





# Ron Clark 16 Copy Mixed Floor Display

**WJ Hussar**





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type of chemical reaction for this reaction we have a neutralization reaction balancing strategies here we have a  
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web the balanced chemical equation for this reaction is  $\text{Mg}(\text{s}) + 2\text{HNO}_3(\text{aq}) \rightarrow \text{Mg}(\text{NO}_3)_2(\text{aq}) + \text{H}_2(\text{g})$  in this equation the s symbolizes  
solid magnesium aq represents aqueous nitric acid aq signifies magnesium nitrate in a dissolved state and g represents the  
gaseous hydrogen produced

$\text{HNO}_3 + \text{Mg}(\text{C}_2\text{H}_3\text{O}_2)_2 \rightarrow \text{H}_2\text{C}_2\text{H}_3\text{O}_2 + \text{Mg}(\text{NO}_3)_2$  chemical equation - Aug 11 2023

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reaction where one mole of solid magnesium mg and two moles of aqueous nitric acid  $\text{HNO}_3$  react to form one mole of  
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