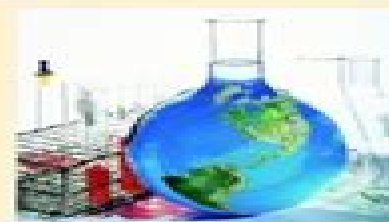


QUÍMICA DEL AGUA. FACTORES QUE AFECTAN A LA CALIDAD DEL AGUA

3.1 INTERACCIONES DE FASES EN EL AGUA: Reconoce las propiedades de las diferentes sustancias sólidas que se encuentran en el agua.

3.2 TRATAMIENTO Y POTABILIZACIÓN DEL AGUA: Describe los procesos químicos que ocurren en las etapas durante el tratamiento de aguas residuales.

3.3 MÉTODOS Y NORMATIVIDAD: Determina la calidad del agua teniendo en cuenta la normatividad vigente en el país.



1. INTERACCIONES DE FASES EN EL AGUA

La mayoría de los fenómenos químicos y bioquímicos importantes en el agua implican interacciones entre las especies que se encuentran en el agua y aquellas que se encuentran en otra fase (sólidos suspendidos o coloides).

Sedimentos en agua

Los sedimentos son capas de materia sólida finamente divididas que cubren los fondos de ríos, arroyos, lagos, embalses, bahías, estuarios y océanos. Los sedimentos típicamente consisten en mezclas de minerales finos, medios y gruesos, incluyendo arcilla, limo y arena, mezclados con materia orgánica. Su composición puede variar desde ser de materia mineral pura a materia predominantemente orgánica.

Los sedimentos son reservorios de una variedad de detritus biológicos, químicos y contaminantes en cuerpos de agua y son depósitos de contaminantes, como metales pesados y compuestos orgánicos tóxicos. De particular preocupación es la transferencia de especies químicas de los sedimentos a la cadena alimenticia acuática de organismos que viven gran parte de su vida en contacto con sedimentos o que viven en ellos. Entre los organismos vivos que viven bajo dichas condiciones se encuentran camarones, cangrejos, gusanos, almejas, etc. La importancia estudiar a los sedimentos radica en estos organismos, dado que se encuentran cerca del base de la cadena alimenticia.

Formación de sedimentos

Los procesos físicos, químicos y biológicos pueden resultar en la deposición de sedimentos en las regiones inferiores de los cuerpos de agua. Los sedimentos pueden formarse por simples erosiones fluviales o también por reacciones de precipitación

Los sedimentos de carbonato de calcio se pueden formar cuando el agua está enriquecida de dióxido de carbono y posee un alto contenido de calcio (dureza), lo que genera una reacción entre ambas y produce dióxido de carbono, el cual es liberado a la atmósfera:



También se puede formar sedimentos de carbonatos de calcio cuando el pH se eleva debido a reacciones fotosintéticas,



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Fabián A. Samaniego



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Table of Contents Selected Chapters From Cuaderno De Traba

1. Understanding the eBook Selected Chapters From Cuaderno De Traba
 - The Rise of Digital Reading Selected Chapters From Cuaderno De Traba
 - Advantages of eBooks Over Traditional Books
2. Identifying Selected Chapters From Cuaderno De Traba
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Selected Chapters From Cuaderno De Traba
 - User-Friendly Interface
4. Exploring eBook Recommendations from Selected Chapters From Cuaderno De Traba
 - Personalized Recommendations

- Selected Chapters From Cuaderno De Traba User Reviews and Ratings
- Selected Chapters From Cuaderno De Traba and Bestseller Lists
- 5. Accessing Selected Chapters From Cuaderno De Traba Free and Paid eBooks
 - Selected Chapters From Cuaderno De Traba Public Domain eBooks
 - Selected Chapters From Cuaderno De Traba eBook Subscription Services
 - Selected Chapters From Cuaderno De Traba Budget-Friendly Options
- 6. Navigating Selected Chapters From Cuaderno De Traba eBook Formats
 - ePub, PDF, MOBI, and More
 - Selected Chapters From Cuaderno De Traba Compatibility with Devices
 - Selected Chapters From Cuaderno De Traba Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Selected Chapters From Cuaderno De Traba
 - Highlighting and Note-Taking Selected Chapters From Cuaderno De Traba
 - Interactive Elements Selected Chapters From Cuaderno De Traba
- 8. Staying Engaged with Selected Chapters From Cuaderno De Traba
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Selected Chapters From Cuaderno De Traba
- 9. Balancing eBooks and Physical Books Selected Chapters From Cuaderno De Traba
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Selected Chapters From Cuaderno De Traba
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Selected Chapters From Cuaderno De Traba
 - Setting Reading Goals Selected Chapters From Cuaderno De Traba
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Selected Chapters From Cuaderno De Traba
 - Fact-Checking eBook Content of Selected Chapters From Cuaderno De Traba

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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