



PRINCIPLES FOR RESPONSIBLE  
MANAGEMENT EDUCATION COLLECTION  
Oliver Laasch, Editor

# **Teaching Ethics Across the Management Curriculum**

*Contributing to a Global  
Paradigm Shift*  
*Volume III*

**Kemi Ogunyemi**

A CRME PUBLICATION

**PRME** Principles for Responsible  
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# School Education Volume 3 Developing A Curriculum

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web 1 if the  $pe$  of the reactants is 430kJ and the  $pe$  of the products is 190kJ what is the  $\Delta H$  2 if the  $pe$  of the reactants is 430kJ and the activation energy is 220 how much total energy will be needed to be added for the reaction to proceed transcribed image text potential energy diagram answer the questions using the graph

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web answer see below explanation the rate of a chemical reaction  $a + b \rightleftharpoons c + d$  is given in first approximation by  $v = k_f [a][b]$  where  $k_f$

where  $k$  depends by many factors as solvent geometry type of reaction  $E_a$  is the activation energy  $R$  is the universal constant of gases  $T$  is the absolute temperature

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