

# READING

## Big Numbers

You know how to read this number:

614

six  
hundred  
fourteen

If you know how to read numbers in the hundreds,  
you can read any large number!

Large numbers are written in **periods** of three numbers.



million

thousand

The periods have names. Simply read the three numbers.  
When you come to a comma, say the name of that period.

Let's give it a try!

6 1 4 , 6 1 4 , 6 1 4

six  
hundred  
fourteen

million

six  
hundred  
fourteen

thousand

six  
hundred  
fourteen

# Reading The Numbers

**AN Whitehead**



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web the exponential law can also be interpreted as the decay probability for a single radioactive particle to decay in the interval  $dt$  about  $t$  this probability  $p(t)$  properly

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web decay 1 parent isotope parent isotope decay radioactive daughter  $t$  radioactive daughter  $t$   $dt$  decay 1 decay 2  $dt$  init radioactive daughter

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web jan 10 2021 although the radioactive decay of a nucleus is too small to see with the naked eye we can indirectly view radioactive decay in an environment called a cloud

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web the decay of radioactivity in a radioactive element can be modelled using cubes dice or coins in decay a radioactive parent nucleus randomly emits an alpha or beta particle

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web this project involved the simulation of radioactive decay the probability  $p$  that a certain atom will decay in a small amount of time  $t$  is  $p = \lambda t$   $1 - \lambda t \approx e^{-\lambda t}$   $\frac{1}{2} = e^{-\lambda t_{1/2}}$  where  $\lambda$

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web the most common types of radioactivity are  $\alpha$  decay  $\beta$  decay  $\gamma$  emission positron emission and electron capture nuclear

reactions also often involve  $\gamma$  rays and some

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web during alpha decay an element emits a helium nucleus with 2 neutrons and 2 protons thus the atomic mass of the new element is decreased by four and the atomic number

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web nov 21 2021 the equation below describes the radioactive decay of a uranium 238 atom where a thorium 234 atom and an alpha particle are formed  ${}^{238}_{92}\text{U} \rightarrow {}^{234}_{90}\text{Th} + {}^4_2\text{He}$

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