Catalysis & Catalysts

Solid Catalysts

□Preparation of catalysts

Precipitation

To form non-soluble precipitate by desired reactions at certain pH and temperature

Adsorption & ion-exchange

Cationic: S-OH+ + C+ → SOC+ + H+

Anionic: S-OH' + A' → SA' + OH'

I-exch. S-Na+ + Ni 2+

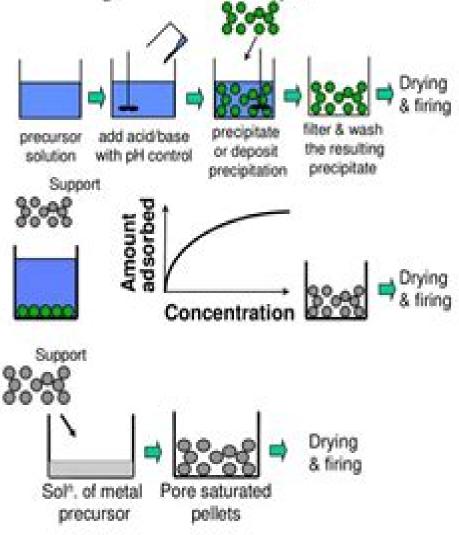
S-Ni 2+ + Na+

Impregnation

Fill the pores of support with a metal salt solution of sufficient concentration to give the correct loading.

Dry mixing

Physically mixed, grind, and fired



Preparation Of Solid Catalysts

Puja Mehta

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