
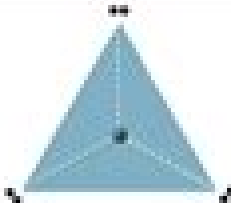
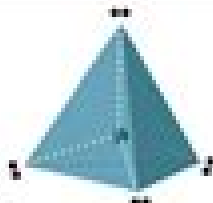
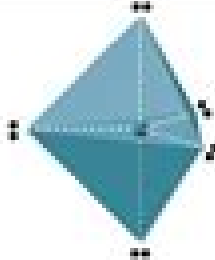




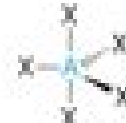
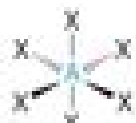



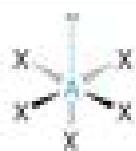
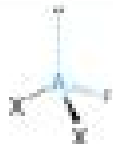

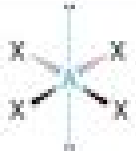



Electron Groups	2	3	4	5	6
Molecular Geometry					
	Linear	Trigonal planar	Tetrahedral	Trigonal bipyramidal	Octahedral
Zero Lone Pairs					
	Linear $AX_2$	Trigonal planar $AX_3$	Tetrahedral $AX_4$	Trigonal bipyramidal $AX_5$	Octahedral $AX_6$
One Lone Pair					
		Bent (V-shaped) $AX_2E$	Trigonal pyramidal $AX_3E$	Seesaw $AX_5E$ One axial lone pair	Square pyramidal $AX_6E$
Two Lone Pairs					
			Bent (V-shaped) $AX_2E_2$	T-shaped $AX_3E_2$ Two axial lone pairs	Square planar $AX_4E_2$
Three Lone Pairs					
				Linear $AX_3E_3$ Three axial lone pairs	

# Shapes And Structures

**M Planty**



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