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# Managing Safety And Health Programs

**Ying Liu**



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web we utilize the equation that relate amount remaining initial mass and number of half lives  $n = \frac{t}{T} \times n_0$  1 2  $n = \frac{t}{T} \times n_0$  1 2 4 x 50 3

125 g 5 what is the half life of an isotope that is 80 remained after 16 days solution remaining 80 100 therefore  $n = \frac{t}{T} \times n_0$  100

now using the half life equation in example 1 we have

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