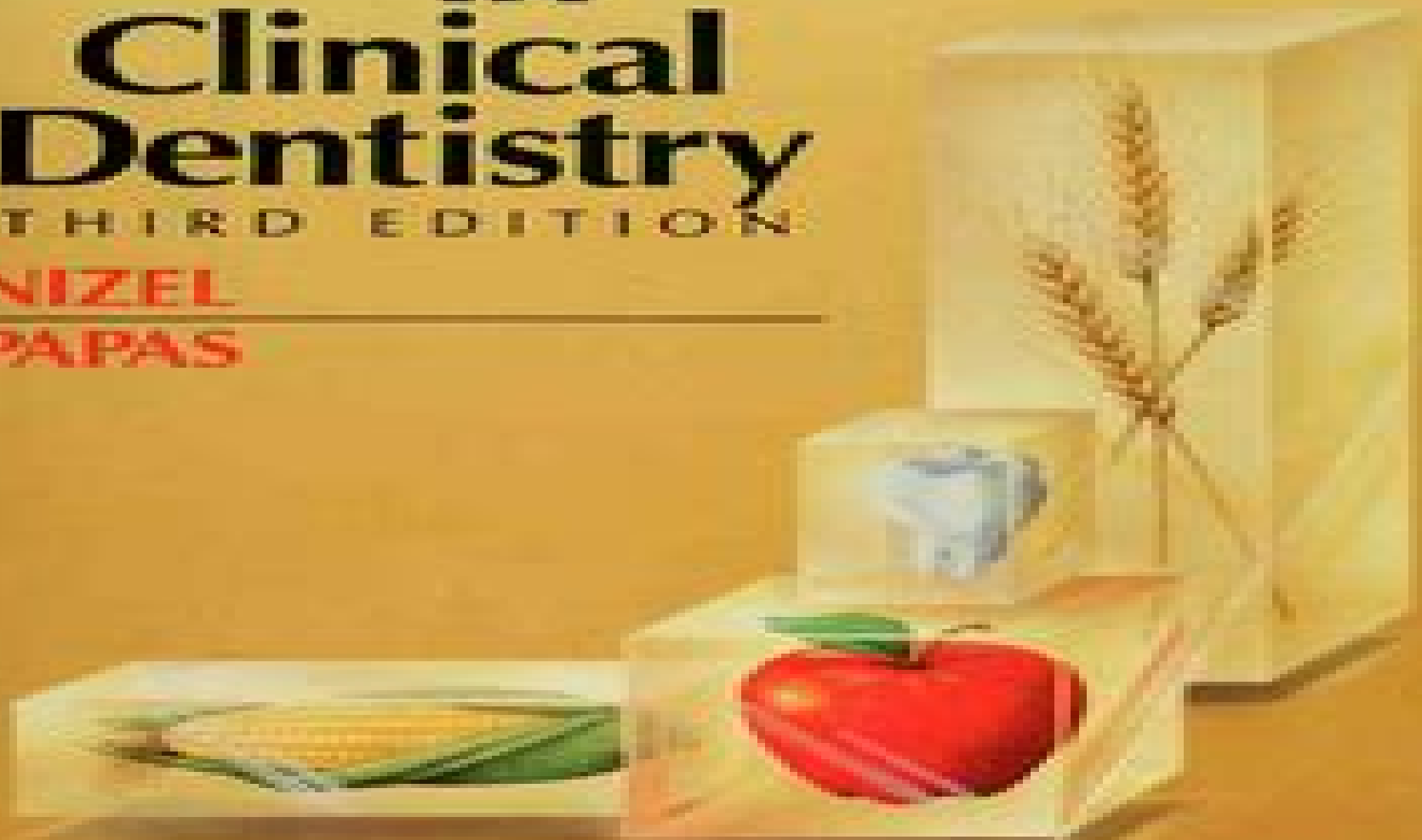


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 $80.0\text{ g I}_2\text{O}_5 \times \frac{1\text{ mol I}_2\text{O}_5}{1\text{ mol I}_2\text{O}_5} \times \frac{1\text{ mol I}_2}{1\text{ mol I}_2\text{O}_5} \times \frac{253.8\text{ g I}_2}{1\text{ mol I}_2} = 28.0\text{ g I}_2$

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