


C. PETER
WAGNER
& PABLO
DEIROS
EDITORS

CONTRIBUTORS:
Carlos Annacondia
Pablo Bottari
Omar Cabrera
Claudio Freidzon
Eduardo Lorenzo
Carlos Mraida
Guillermo Prein
Jane Rumph
Ed Silvoso
Juan Zuccarelli



THE RISING REVIVAL

FIRSTHAND ACCOUNTS OF THE INCREDIBLE
ARGENTINE REVIVAL—AND HOW IT
CAN SPREAD THROUGHOUT THE WORLD

Rising Revival What The Spirit Is Saying Through The Argentine Revival

Annelies Wilder-Smith



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end point and that value was used in unit analysis to find the molar mass and identify the unknown solid acid

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web by utilizing the half equivalence point on the titration curve the pka of the acid analyte was determined to be 3.655 furthermore using the stoichiometric ratio between the naoh and the acid analyte 1:1 we could experimentally determine the number of moles of the acid in the solution to be 0.00731 moles making the experimental molar mass

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web goals understand that parts of an acid base titration be able to determine the K_a or K_b from pH data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate K_{a1} and K_{a2} for a polyprotic acid by the end of this lab students should be able to

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 $HCl(aq) + NaOH(aq) \rightarrow NaCl(aq) + H_2O(l)$
one mole of HCl would be fully neutralized by one mole of NaOH

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solution of known strength is added to a certain volume of a treated sample containing an

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web conclusion this lab evaluated an acid base reaction with the volumetric analysis of a standard solution being naoh reaction with a 2nd substance $\text{H}_2\text{C}_2\text{O}_4$ aq 2NaOH aq $\text{Na}_2\text{C}_2\text{O}_4$ $4\text{H}_2\text{O}$ to differentiate an acid from a base in a solution a ph scale is used the ph scale ranges from 0 to 14 a ph of 7 is neutral in an acid base reaction

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