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Methods for Physical & Chemical Analysis of Fresh Waters

H L Golterman R S Clyma
M A M Obstade

$$\frac{1}{2}\text{Ca}^{2+} = 2.68$$



$$\frac{1}{2}\text{Mg}^{2+} = 0.74$$

$$\text{NH}_4^+ = 0.05$$



$$\text{K}^+ + \text{Na}^+ = 0.05$$

$$\text{Total} = 4.12 \text{ mmol L}^{-1}$$

$$\text{HCO}_3^- = 3.54$$

$$\frac{1}{2}\text{SO}_4^{2-}$$



$$\text{Cl}^- = 0.48$$

$$\text{PO}_4\text{-P} = 0.01$$



$$= 4.15 \text{ mmol L}^{-1}$$

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