

**AIR, COMBUSTION, RUSTING AND FIRE FIGHTING**

1. Extinguishing a fire caused by burning kerosene, carbon dioxide is used in preference to water. Explain why?
2. Give a reason why, carbon dioxide is used as a fire extinguisher?
3. Carbon dioxide is not used as a fire extinguisher in class D fires. Explain why?
4. Burning magnesium continue to burn inside the gas jar full of carbon dioxide. Explain why?
5. A piece of burning magnesium continues to burn in a gas jar of Sulphur dioxide gas. Explain why?
6. A small child has been playing with matches and its clothes have caught fire. Explain how you would put out the flames.
7. Zinc is used to cover iron so as to prevent rusting of iron. Explain why?
8. Explain why motor vehicles based in Mombasa rust much faster than those based in Nairobi?
9. If the tin surface is scratched then the iron will rust. Why does this not happen with zinc plating (galvanizing)?
10. Aluminium metal would protect iron from rusting better than zinc metal. Explain why?
11. It is not advisable to wash vehicles using sea water. Explain why?
12. Why do car exhaust rust faster than other structures made of steel?
13. The parts of shipwrecks made of iron rust more slowly in deep water. Explain why?
14. Alkaline medium inhibits the rusting of iron. Explain why?
15. Iron does not rust even if zinc coating is broken in a galvanized iron pipe. Explain why?
16. Zinc is used to coat for iron and not vice versa. Explain why?

17. A sample of iron weighs more after it rusts. Explain why?
18. The percent of carbon dioxide in the atmosphere does not significantly vary from 0.04%. Explain why?
19. The air over an open flame is always moist. Explain why?
20. Explain why, cooking pots made up of aluminium do not corrode easily when exposed to air?
21. Why does a magnesium ribbon be cleaned before burning in air?
22. When a candle was burnt completely, the total mass of the products was found to be greater than the original mass of the candle. Explain why?
23. In temperate countries, salt is sprayed on the roads to defrost and clear roads but the long-term effect on this practice is costly to motorists. Explain why?
24. Why baking soda does extinguish fire?
25. Tetrachloromethane is used as a fire extinguisher but not carbon disulphide. Explain why?
26. Explain why magnesium gains in mass but coal loses mass when each is burned in air?

**HEAT SOURCES AND FLAMES**

1. When air holes of a Bunsen burner are fully opened the Bunsen-burner produce a non-luminous flame. Explain why?
2. It is advisable to set the Bunsen burner to luminous flame prior to experiment. Explain why?
3. It is advisable to turn a non-luminous flame to luminous flame while the flame is not in use. Explain why?
4. Non-luminous flame is used for heating. Explain why by giving two reasons?
5. Explain how hotness of Bunsen burner flame can be increased.

# Reasoning In Chemistry Pb 1995

**Tao Wei**



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## **Table of Contents Reasoning In Chemistry Pb 1995**

1. Understanding the eBook Reasoning In Chemistry Pb 1995
  - The Rise of Digital Reading Reasoning In Chemistry Pb 1995
  - Advantages of eBooks Over Traditional Books
2. Identifying Reasoning In Chemistry Pb 1995
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Reasoning In Chemistry Pb 1995
  - User-Friendly Interface
4. Exploring eBook Recommendations from Reasoning In Chemistry Pb 1995
  - Personalized Recommendations
  - Reasoning In Chemistry Pb 1995 User Reviews and Ratings
  - Reasoning In Chemistry Pb 1995 and Bestseller Lists
5. Accessing Reasoning In Chemistry Pb 1995 Free and Paid eBooks
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  - Reasoning In Chemistry Pb 1995 eBook Subscription Services

- Reasoning In Chemistry Pb 1995 Budget-Friendly Options
- 6. Navigating Reasoning In Chemistry Pb 1995 eBook Formats
  - ePub, PDF, MOBI, and More
  - Reasoning In Chemistry Pb 1995 Compatibility with Devices
  - Reasoning In Chemistry Pb 1995 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Reasoning In Chemistry Pb 1995
  - Highlighting and Note-Taking Reasoning In Chemistry Pb 1995
  - Interactive Elements Reasoning In Chemistry Pb 1995
- 8. Staying Engaged with Reasoning In Chemistry Pb 1995
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Reasoning In Chemistry Pb 1995
- 9. Balancing eBooks and Physical Books Reasoning In Chemistry Pb 1995
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Reasoning In Chemistry Pb 1995
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Reasoning In Chemistry Pb 1995
  - Setting Reading Goals Reasoning In Chemistry Pb 1995
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Reasoning In Chemistry Pb 1995
  - Fact-Checking eBook Content of Reasoning In Chemistry Pb 1995
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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