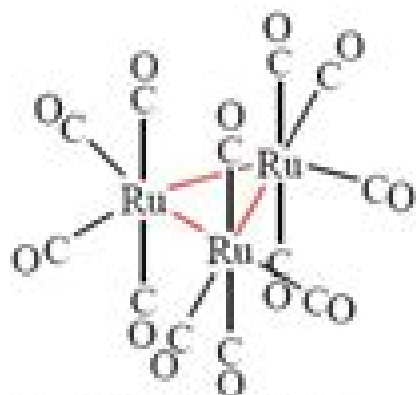
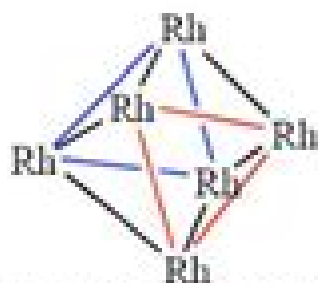


Small metal clusters

The simplest metal cluster contains the M₃ unit, held together by metal-metal bonds (rather than solely by bridging ligands). An example is Ru₃(CO)₁₂. The trigonal arrangement of metal atoms is reminiscent of the close-packing of spheres in bulk metal. The cluster Rh₆(CO)₁₆ has an octahedral arrangement of metal atoms (with 12 terminal CO ligands and 4 triply-bridging CO ligands). The metal organization is analogous to the packing of two trigonal M₃ units in adjacent layers in bulk metal.



structure of Ru₃(CO)₁₂



framework of Rh₆(CO)₁₆



Metal Clusters In Chemistry

H Kauffman



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