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web we can write a balanced nuclear equation for this reaction as let s do a practice example identify the product x by writing a balanced nuclear equation ${}^{137}_{55}\text{Cs} \rightarrow {}^{137}_{56}\text{Ba} + x$ because the mass number is the same on both sides of the equation x must have a

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web identify the missing particle in the following nuclear reaction ${}^1_0\text{n} + {}^{235}_{92}\text{U} \rightarrow {}^{137}_{52}\text{Te} + {}^{97}_{40}\text{Zr} + 2\text{He} + 3\text{Sr} + 6\text{Zr} + 40\text{Zr}$

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