

## ARRT Radiography Exam (Mosby Review) With A Guaranteed A+

**Thermal energy** - Kinetic heat energy resulting from the movement of atoms or molecules.

**Nuclear energy** - Kinetic energy from the nucleus of an atom.

**Electromagnetic energy** - Kinetic energy that is carried and transferred through fields.

**Velocity (speed)** - Kinetic force that an object is moving.

**Acceleration** - Kinetic the rate of change of speed per unit of time.

**Work** - Kinetic force applied on an object over a distance.

**Power** - Kinetic the rate of doing work (measured in watts).

**Atomic number** - Kinetic number of protons (positive charges) and neutrons (no charge) in the core of the nuclei of an atom.

**Atomic mass** - Kinetic number of protons plus number of neutrons, represented by the letter A.

**Electron shells** - Kinetic number of electron shells (negative charges) electron shells represented by the letters K, L, M, N, O, P, and Q, or 1, 2, 3, 4, 5, 6, and 7, or in Roman numerals. The number of electrons and protons is equal.

**Atomic number of an atom** - Kinetic equals the number of protons in the nucleus, represented by the letter Z. The atomic number determines the chemical elements, all chemical elements are represented in the periodic table of the elements.

**Isotopes** - Kinetic atoms with the same number of protons but with a different number of neutrons.

**Electromagnetic energy** - Kinetic force that holds electrons in orbit around the nucleus.

**Orbit rule** - Kinetic the outer shell of an atom may not contain more than eight electrons.

**Photoelectric radiation** - Kinetic alpha particles (helium nucleus) two protons and two neutrons, beta particles (electron) the particles emitted from the nucleus of a radioactive atom.

**Photon** - Kinetic the smallest amount of any type of electromagnetic radiation, also considered a bundle of energy called a quantum, travels at the speed of light, travels in waves or a straight path.

# Mosbys General Rx

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