



Radical Optimism

Frederick C Teiwes



Radical Optimism:

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web balancing equations the chemical equation described in section 4 1 is balanced meaning that equal numbers of atoms for each element involved in the reaction are represented on the reactant and product sides this is a requirement the equation must satisfy to be consistent with the law of conservation of matter

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web a key step in balancing chemical equations is correctly identifying the formulas of the reactants and products for example consider the reaction between calcium oxide CaO and H_2O to form aqueous calcium hydroxide Ca(OH)_2 is it possible to balance the equation if you incorrectly identify the product as CaOH and if so what is the equation

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