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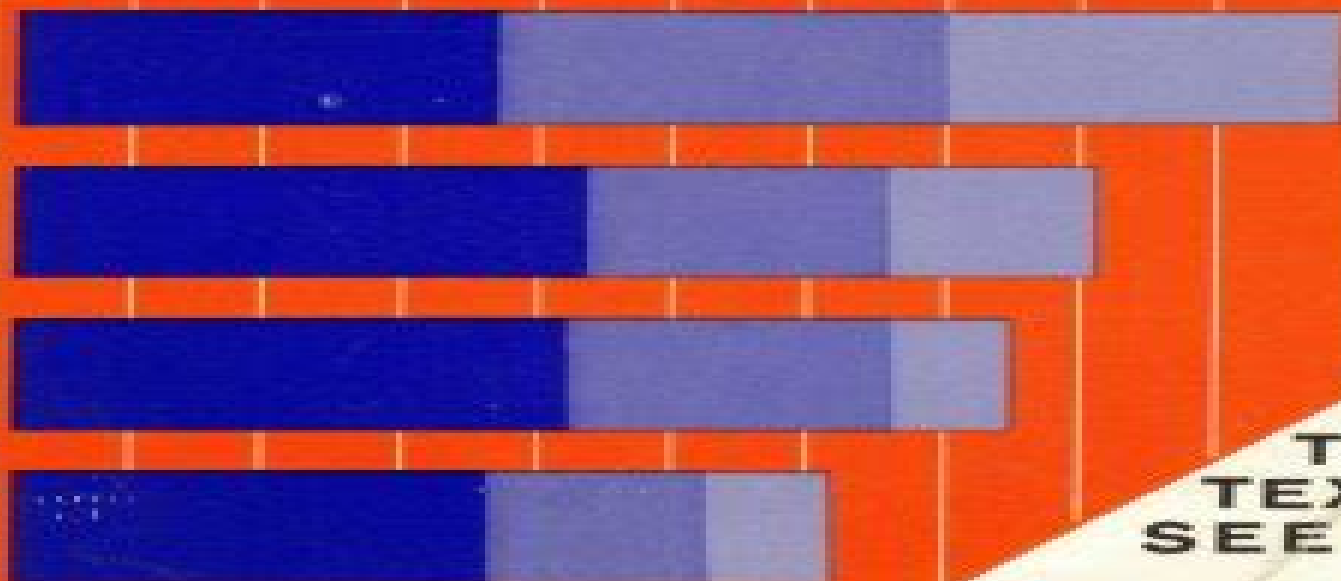
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despite his massive military fearsome security police and superpower support was overthrown by a popular and largely peaceful revolution But the CIA was not alone in its myopia as Charles Kurzman reveals in this penetrating work Iranians themselves except for a tiny minority considered a revolution inconceivable until it actually occurred Revisiting the circumstances surrounding the fall of the shah Kurzman offers rare insight into the nature and evolution of the Iranian revolution and into the ultimate unpredictability of protest movements in general As one Iranian recalls The future was up in the air Through interviews and eyewitness accounts declassified security documents and underground pamphlets Kurzman documents the overwhelming sense of confusion that gripped pre revolutionary Iran and that characterizes major protest movements His book provides a striking picture of the chaotic conditions under which Iranians acted participating in protest only when they expected others to do so too the process approaching critical mass in unforeseen and unforeseeable ways Only when large numbers of Iranians began to think the unthinkable in the words of the U S ambassador did revolutionary expectations become a self fulfilling prophecy A corrective to 20 20 hindsight this book reveals shortcomings of analyses that make the Iranian revolution or any major protest movement seem inevitable in retrospect

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each compound to balance $\text{Mg} + \text{HNO}_3 \rightarrow \text{Mg}(\text{NO}_3)_2 + \text{NO}_2 + \text{H}_2\text{O}$ you'll need to be

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$\text{Mg}^{2+}(\text{aq}) + 2\text{OH}^{-}(\text{aq}) + 2\text{H}^{+}(\text{aq}) + 2\text{NO}_3^{-}(\text{aq}) \rightarrow \text{Mg}^{2+}(\text{aq}) + 2\text{NO}_3^{-}(\text{aq}) + 2\text{H}_2\text{O}(\text{l})$ net ionic equation $2\text{OH}^{-}(\text{aq}) + 2\text{H}^{+}(\text{aq}) \rightarrow 2\text{H}_2\text{O}(\text{l})$ sulfuric acid and lithium hydroxide

balanced total equation $\text{H}_2\text{SO}_4(\text{aq}) + 2\text{LiOH}(\text{aq}) \rightarrow \text{Li}_2\text{SO}_4(\text{aq}) + 2\text{H}_2\text{O}(\text{l})$