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First results on radiometric dating of metals by alpha spectrometry :



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ABSTEACT

A new TSAC (Thick Source Alpha Particle-Counting spectrometry) setup of high prosected efficiency and low intrinsic background was half to measure the alpha particles coming from the ²⁻⁸U-and ²⁻⁸Th decay chaims such exactions show no the permitted as such exactions to be a the ²⁻⁸Ra for the entry chaims such confly direct metal during method proposed to date. Micro-POU, micro-RES and SESA-EDS were used as complementary analytical techniques in order to search for heterogenesties and/or impurities that could bin the TSAC state. Once, resulting slags and metallic prills from two recent uneiting experiments that reproduced are instructed niques and two ancient coins twen stadied to determine how accondities fractions from occurs tharing smelling. With this TSAC setup the background's grad was extremely love, but comparable to the signal from the measured metals. Preliminary results show no evidence of a preferential depletion of ²⁰⁰Ra in ameliac inetal with on post to the original from the original form produced metals.

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1. Introduction

Archaeometallurgical research focuses in the reconstruction of early trade routes in terms of location of metal sources (ore deposits) and in the study of metallic artifacts and metallurgical remains through various analytical and examination techniques. Important information can be derived from these studies, like ore provenance, contact between different ancient societies and their technological development.

It's not an overstatement to say that the Holy Grail of archaeometallurgists is to achieve a method that can directly give the date of the manufacture of a metallic artifact.

Until now, the dating of metallurgical processes is made indirectly, under favor able conditions, by [1]; a) thermoluminescence (TL) and optical stimulated luminescence (CSL) of burnt soil in kilns and cracibles, or of remaining over material in the metal artifact and slag products [2]; b) radiocurbon (**C) applied on carbon found in trupped slag inclusions (mainly iron a rtifacts); c) as sociation with other archeological artifacts found in the same archeological context; and d) some aging or degradation specificities that may differentiate an old from a new object (e.g. interpretallar corrosion is a slower corrosion process).

In 2006 [1] I. Liritzis proposed a radiometric dating method for metals that are produced from ones (native metals excluded), based on the isotopic fractionation that may occur during one smelting. This fractionation disrupts the secular equilibrium in the ²³⁶U series. In this sense, ²³⁶Ra, a radiomodide that occurs naturally in the decay chain of ²³⁶U with a half-life of 1602 years, tends to be withdrawn by slag products away of the melted metal, as are salt of its low specific gravity. 5–6 g/cm³, and its lower melting point (700 °C) when compared to some metals, as for example copper (8.9 g/cm³: 1083 °C).

As a consequence, in a metal, ²⁰⁰Ra would appear from the decay of its parent. ²¹⁰Th (half-life, t_{1,0} = 80000 years). Assuming that initially in the metal ²²⁰Ra is not present, then at time t, the ratio between the activities of ²²⁰Ra and ²²⁰Th is given by Eq. (1) [1]

$$\frac{A_{\text{SSMs}}}{A_{\text{SSMs}}} = \frac{A_{\text{SSMs}}}{\lambda_{\text{SSMs}} - \lambda_{\text{SSMs}}} \left[1 - e^{-i\lambda_{\text{SSMs}} - \lambda_{\text{SSMs}}}\right],$$
 (1)

where $\lambda_{\rm COLO_{1}}$ and $\lambda_{\rm COLO_{2}}$ are the decay constants ($\lambda = \ln(2)/\Gamma_{\rm LO}$) of $^{23.6}{\rm Ra}$ and $^{2.66}{\rm Th}$, respectively. As $\lambda_{\rm COLO_{2}} > \lambda_{\rm COLO_{2}}$ (4.3.3 × 10 $^{-6}{\rm yr}^{-1} > \infty$ 8.56 × 10 $^{-6}{\rm yr}^{-1}$) the ratio of activities tend to the value $\lambda_{\rm COLO_{2}} / (\lambda_{\rm COLO_{2}} / \lambda_{\rm COLO_{2}} / \lambda_{\rm$

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