

# Reading Abstract Expressionism

## Context and Critique

Edited and with an Introduction by Ellen G. Landau



# Reading Abstract Expressionism Context And Critique

**JE Gale**



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web during this reaction the  $\text{H}^+$  ions from the acid and the  $\text{OH}^-$  ions from the alkali react together to form water as this happens energy is released which is detected by the thermometer

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measure an appropriate volume of each liquid eg 25 cm<sup>3</sup> place one of the

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web for reactions involving acetic acid or ammonia the measured enthalpy change of neutralization is a few kJ less  
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eg 1 HCl(aq) + NaOH(aq) → NaCl(aq) + H<sub>2</sub>O(l) for which ΔH = 57 kJ mol<sup>-1</sup> eg 2 HNO<sub>3</sub>(aq) + KOH(aq) → KNO<sub>3</sub>(aq) + H<sub>2</sub>O(l) for which ΔH = 57 kJ mol<sup>-1</sup>

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web the enthalpy change of neutralisation is defined as the enthalpy change when solutions of an acid and alkali react together to produce one mole of water for example  $\text{hcl aq} + \text{naoh aq} \rightarrow \text{nacl aq} + \text{h}_2\text{o aq}$  you can see that one mole of water is produced when one mole of hcl or one mole of naoh reacts

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