#### MATTER AND ITS CHANGES

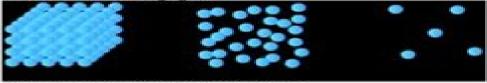
#### \*EFFECT OF CHANGE IN TEMPERATURE ON THE STATES OF MATTER

We know that matter exists in three different physical states – solid, liquid, and gas. The states of matter are interchangeable i.e., a substance can be changed from its solid state to its liquid state and vice-versa. Transitions from the liquid state to the gaseous state and vice-versa are also possible. However, some substances directly change into their gaseous states from their solid states and vice-versa.

Water exists in all three states (i.e., solid, liquid, and gaseous) as ice, water, and water vapour respectively. These states are interchangeable (see the given figure).



Temperature is a major factor responsible for causing a change in the various states of matter. The changes in the states of matter with increasing or decreasing temperatures can be shown in the form of the following figure.



Effect of temperature on the states of matter

On heating a solid substance, the kinetic energy of its constituent particles increases. As a result, the particles start vibrating with greater speed. This extra energy helps the particles to overcome the inter-particle force of attraction. Soon, they leave their positions and start moving more freely. Consequently, the substance melts. The change of solid state into liquid state is known as fusion. The minimum temperature at which a substance starts melting and becomes a liquid at atmospheric pressure is known as its melting point.

It has been observed that temperature remains constant till all solid melts into liquid. The heat supplied is used for changing a solid state into its liquid state by overcoming the particle-particle attraction force. The solid substance absorbs heat energy without showing any rise in temperature. This heat energy is called latent heat. The amount of heat required to convert a unit mass of a solid into its liquid state, without a change in temperature (i.e., at its melting point), is called the latent heat of fusion.

For ice, the latent heat of fusion is 334 kJkg-1. This means that 334 kJ of heat is required to convert 1 kg of ice at 0°C into 1 kg of water at 0°C. Conversely, 334 kJ of heat is released when 1 kg of water freezes at 0°C to give 1 kg of ice at 0°C.

If heating is continued, then the kinetic energy of liquid particles increases further. This increases the velocity of the particles, and at a certain temperature, they obtain enough energy to break free from the force of attraction of the other particles. At this point, the liquid starts boiling i.e., starts

# **Matter And Its Changes**

James E. Brady,Frederick Senese,Neil D. Jespersen

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Latin American Perspectives on Women Philosophers in Modern History Christine Lopes, Katarina Ribeiro Peixoto, Pedro Pricladnitzky, 2022-06-21 This book presents Latin American Perspectives on women philosophers comprising selected articles from the First International Conference of Women in Modern Philosophy that took place in Rio de Janeiro City Brazil Latin America in June of 2019 The conference brought together over twenty national transnational and international philosophers from seven countries whose work combines historical and analytical insight to recover the philosophical legacy of women philosophers Historical and analytical work on women sphilosophical thought constitute efforts to re conceptualize what counts as philosophical knowledge and re appraise the epistemic relevance of written material that women thinkers produced for most of history This collection and the conference that gave origin to it are testimony to the enduring power of multinational and multicultural philosophical collaboration NJ Ask: Science, Grade 4 Lauren Fletcher, Amy Konzelmann, 2012-02-24 All fourth grade students in NJ are required to pass the NJ ASK Assessment of Skills and Knowledge Grade 4 Science assessment test REA s test prep gives fourth graders all the information they need to succeed on this important high stakes exam Completely aligned with the core curriculum standards of the NJ Department of Education the test prep includes a student friendly targeted review of the science skills tested on the exam including life science physical science and earth science Our focused lessons appeal to students at all learning levels Each lesson explains science topics in language suitable for the fourth grade level while numerous drills strengthen abilities Color icons throughout the book highlight important questions and study tips The book also includes two full length practice tests with detailed explanations of answers that allow students to test their knowledge and focus on areas in need of improvement Samuel Butler Samuel Butler, 2019-11-21 The Note Books of Samuel Butler is an intriguing compilation that offers profound insights into Butler's intellectual musings and philosophical reflections on a diverse range of subjects including art science and society The literary style is characterized by Butler's sharp wit and keen observational prowess often presented in the form of aphorisms and succinct commentary which invites readers to engage critically with his thoughts This collection acts as both an autobiographical window into the mind of a 19th century Victorian intellectual and a broader commentary on his contemporary society revealing the tensions between tradition and modernity that permeate his work Samuel Butler was a polymath a novelist essayist and social critic known for his iconoclastic ideas especially towards Darwinism and Victorian values His discontent with societal norms and traditional beliefs stemming from his own unique experiences including his stay in New Zealand shaped his unconventional worldview These influences are palpable in his note books as he navigates both personal introspection and societal critique melding philosophies and experiences into a rich tapestry of thought This illuminating collection is highly recommended for readers interested in the interplay of art philosophy and social critique during the Victorian era Butler's notes not only reflect his creative genius but also serve as a thought provoking commentary that remains relevant in today s conversations about individualism and societal expectations Handbook of Water and

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