



# THE METRO POEMS

Peter Hughes

# Metro Poems Rome 198891

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web a similar unit of concentration is molality m which is defined as the number of moles of solute per kilogram of solvent not per liter of solution molality =  $\frac{\text{moles solute}}{\text{kilograms solvent}}$  mathematical manipulation of molality is the

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web 1 molarity =  $\frac{\text{mass}}{\text{molar mass}} \times \frac{1000}{\text{volume}}$  0.4500 l 0.825 g 141.9579 g/mol  $\times \frac{1}{0.0129 \text{ m}^3}$  molality =  $\frac{0.825 \text{ g}}{141.9579 \text{ g/mol}} \times \frac{1000}{0.00581158 \text{ mol}} \times \frac{1}{0.00581158 \text{ mol}} \times \frac{1}{0.4500 \text{ kg}} \times \frac{1}{0.0129 \text{ m}^3}$  mole fraction =  $\frac{2 \text{ hpo}_4}{0.825 \text{ g} + 141.9579 \text{ g/mol} \times 0.00581158 \text{ mol}} \times \frac{1}{450.0 \text{ g} + 18.015 \text{ g/mol} \times 24.97918401 \text{ mol}}$

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$$\frac{\text{mass of a component}}{\text{mass of component in solution total mass of solution}} \times 100$$
 end array

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