

# **Relatively Royal**

**Royal Physical Society of Edinburgh** 

## **Relatively Royal:**

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The Annual of the Royal School of Naval Architecture and Marine Engineering Anonymous,2022-11-04 Reprint of the original first published in 1871

Proceedings of the Royal Society of London Royal Society (Great Britain),1902

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the aldol condensation is extremely important because it can form a  $\boldsymbol{\beta}$  hydroxy aldehyde or

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web an aldol reaction was performed by combining acetone and benzaldehyde to make dibenzalacetone the percent yield of dibenzalacetone was found to be 73 which is a good percent yield the melting point of dibenzalacetone was around the range of 104 110 c this helped conclude that dibenzalacetone had a trans trans e e isomer

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produce an a b unsaturated aldehyde and water figure 1 aldol condensation of ethanal to

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