GCSE Chemistry (AQA Higher Tier) Multiple Choice Questions

Unit: C1

| Q# | Question | A | 8 | C | D |
|----|---|------------|------------|------------|------------|
| 1 | Which of these is the correct group number for Magnesium? | 1 | 2 | 3 | 4 |
| 2 | Which of these elements forms a 1+ ion when bonded ionically? | Mg | Na | РЬ | AJ |
| 3 | Which of these elements is in Group 3? | Al | Pb | Fe | Br |
| 4 | Which of these is the correct periodic table symbol for Gold? | Gu | Go | Au | Ag |
| 5 | Which ion is formed by Aluminium when it bonds ionically? | 1+ | 2+ | 3+ | 4+ |
| 6 | Which of these elements is liquid at room temperature? | Al | Fe | He | Br |
| 7 | Which group contains a series of unreactive gases? | 4 | 7 | 8/0 | 3 |
| 8 | Which of these elements, despite being in Group 1, does not have all the qualities of a metal? | He | н | Hg | Te |
| 9 | Which of these is the correct mass and charge of a neutron? | M:1 C:0 | M:1 C+1 | M:0 C:1 | M:0 C+1 |
| 10 | Three neutrons, two protons and one electron would have a cumulative charge equal to which of these? | +1 | -1 | +2 | -2 |
| 11 | Which of these is the correct approximate mass of an electron? | 1/1200 | 1/1800 | 1/290 | 1/10 |

Questions In Higher Chemistry

Douglas Buchanan, James Melrose

Questions In Higher Chemistry:

Revision Questions for Higher Chemistry Douglas Buchanan, James Melrose, 2015 **Answers to Revision** Ouestions for Higher Chemistry ,2011 Essential SOA Exam Practice: Higher Chemistry Questions and Papers Barry McBride, 2019-12-02 Exam board SQA Level Higher Subject Chemistry First teaching September 2018 First exam Summer 2019 Practice makes permanent Feel confident and prepared for the SQA Higher Chemistry exam with this two in one book containing practice questions for every question type and topic plus two full practice papers Choose to revise by question type or topic A simple grid enables you to pick particular question styles or course areas that you want to focus on with answers provided at the back of the book Understand what the examiner is looking for Clear guidance on how to answer each question type is followed by plenty of questions so you can put the advice into practice building essential exam skills Remember more in your exam Repeated and extended practice will give you a secure knowledge of the key areas of the course chemical changes and structure nature s chemistry chemistry in society researching chemistry Familiarise yourself with the exam papers Both practice papers mirror the language and layout of the real SQA papers complete them in timed exam style conditions to increase your confidence before the exams Find out how to achieve a better grade Answers to the practice papers have commentaries for each guestion with tips on writing successful answers and avoiding common mistakes Fully up to date with SOA's requirements The questions mark schemes and guidance in this practice book match the requirements of the revised SQA Higher Chemistry specification for examination from 2019 onwards Answers to **Revision Questions for Higher Chemistry** Buchanan,2000-12-01 Revision Ouestions for Higher Chemistry Douglas A. Buchanan, James R. Melrose, 1999-04-01 **Answers to Revision Questions for Higher Chemistry** Douglas Buchanan, James Melrose, A. R. Ross, 2015 Higher Chemistry, Second Edition John Anderson, 2019-06-10 Exam Board SQA Level Higher Subject Chemistry First Teaching August 2018 First Exam June 2019 Full course coverage in this new Higher Chemistry textbook updated for the latest changes to the SQA coursework and question papers In text Study and End of course questions have been updated and extended in this edition testing students knowledge and understanding of the chemistry presented and offering lots of practice to revise and consolidate ahead of the exam Worked examples show common Higher Chemistry questions and ways of answering that cover the necessary points Checklists for Revision provide short summaries of the key learning points at the end of each chapter so that students can this to self check their learning which helps them revise for assessments Brand new section Additional features of the Higher Chemistry exam offers advice on how to tackle two important new features of the exam Numeracy and Open ended questions Key terms and Chemical Dictionary aid understanding and allow students to check their knowledge of the key terms **How to Pass Higher** Chemistry for CfE John Anderson, 2014-10-31 Exam Board SQA Level Higher Subject Chemistry First Teaching September 2014 First Exam Summer 2015 Endorsed by SQA Get your best grade with this guide to Higher Chemistry for CfE This book

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