

PERIODIC TABLE

Atomic Properties of the Elements

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National Institute of
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U.S. Department of Commerce

		Frequently used fundamental physical constants										Physics Laboratory		Standard Reference Data						
		For the most accurate values of these and other constants, visit physics.nist.gov/constants										physics.nist.gov		www.nist.gov/srd						
		1 second = 9 192 631 770 periods of radiation corresponding to the transition between the two hyperfine levels of the ground state of ¹³³ Cs																		
		speed of light in vacuum c 299 792 458 m s ⁻¹ (exact)																		
		Planck constant h 6.626 069 3 × 10 ⁻³⁴ J s (exact)																		
		elementary charge e 1.602 176 634 × 10 ⁻¹⁹ C																		
		electron mass m _e 9.109 382 15 × 10 ⁻³¹ kg																		
		proton mass m _p 1.672 621 63 × 10 ⁻²⁷ kg																		
		fine-structure constant α 1/137.035 999 074																		
		Rydberg constant R _∞ 10 973 731.568 16 × 10 ¹⁰ m ⁻¹																		
		Boltzmann constant k 1.380 658 × 10 ⁻²³ J K ⁻¹																		

Atomic Number
 Symbol
 Name
 Atomic Weight
 Ground-state Configuration
 Ionization Energy (eV)

58
Ce
 Cerium
 140.116
 [Xe]4f¹5d¹6s²
 5.5387

*Based upon ^{12}C . () indicates the mass number of the longest-lived isotope.

For a description of the data, visit physics.nist.gov/data

NIST SP 966 (September 2010)

Periodic Table Of The Elements

Mary Griffin



Periodic Table Of The Elements:

Mystery of the Periodic Table Benjamin D Wiker,2003-04-18 Leads the reader on a delightful and absorbing journey through the ages on the trail of the elements of the Periodic Table as we know them today He introduces the young reader to people like Von Helmont Boyle Stahl Priestly Cavendish Lavoisier and many others all incredibly diverse in personality and approach who have laid the groundwork for a search that is still unfolding to this day The first part of Wiker s witty and solidly instructive presentation is most suitable to middle school age while the later chapters are designed for ages 12 13 and up with a final chapter somewhat more advanced Illustrated by Jeanne Bendick and Ted Schluenderfritz

The Periodic Table Adrian Dingle,Simon Basher,2007 A mixture of science and art creates a periodic table never seen before Periodic Table, The: Past, Present, And Future Geoff Rayner-canham,2020-08-04 This is an absolutely wonderful book that is full of

gems about the elements and the periodic table All in all the book is highly recommended to philosophers of chemistry As philosophers we have a natural tendency to concentrate on generalities and not to get too involved in the specifics and the details Above all else this new book reminds us that such an approach needs to be tempered by a detailed knowledge of the exceptions and features that go against the simplified generalities which we so cherish Read Full Review Eric

ScerriFoundations of Chemistry Many questions are dealt with in a clearly written way in this stimulating and innovative book The reader will quickly become interested in the subject and will be taken on tour through this Periodic Table in a very readable way both for students and teachers The number of illustrations is good and clear This book is indeed unique and quite thought provoking This book is highly recommended for students teachers researchers and not only chemists

Geologists biochemist and also physicists will find it very interesting to read Read Full Review Chemistry InternationalThat fossilized chart on every classroom wall isn t that The Periodic Table Isn t that what Mendel ev devised about a century ago No and No There are many ways of organizing the chemical elements some of which are thought provoking and which reveal philosophical challenges Where does hydrogen belong Can an element occupy more than one location on the chart Which are the Group 3 elements Is aluminum in the wrong place Why is silver I like thallium I Why is vanadium like molybdenum Why does gold form an auride ion like a halide ion Does an atom know if it is a non metal or metal Which elements are the metalloids Which are the triels So many questions In this stimulating and innovative book the Reader will be taken on a voyage from the past to the present to the future of the Periodic Table This book is unique This book is readable This book is thought provoking It is a multi dimensional examination of patterns and trends among the chemical elements Every reader will discover something about the chemical elements which will provoke thought and a new appreciation as to how the elements relate together *Periodic Table of the Elements* Ekkehard Fluck,Klaus G. Heumann,1986-07-09

Exploring the Elements Isabel Thomas,2020 Science meets design in this comprehensive introduction to the chemical elements that make up our universe **The Periodic Table** Tom Jackson,2020-02-04 Which is the densest element Which has the largest

atoms And why are some elements radioactive From the little known uses of gold in medicine to the development of the hydrogen bomb this is a fresh new look at the Periodic Table Combining cutting edge science with fascinating facts and stunning infographics this book looks at the extraordinary stories of discovery amazing properties and surprising uses of each elements whether solid liquid or gas naturally occurring synthesised or theoretical From hydrogen to oganesson this is a fact filled visual guide to each element each accompanied by technical data category atomic number weight boiling point as well as fun facts and stories about their discovery and surprising uses The Periodic Table Mary Griffin,2018-12-15 The periodic table is the way scientists have organized the known elements but up and coming scientists can learn much from the periodic table as well Your readers will learn how to read the periodic table including what an element s atomic number means and what the rows and columns signify They ll gather facts about common and interesting elements and even discover how they can predict how elements might chemically react from their position in the table Fact boxes and images add additional information to the accessible text *The Basics of the Periodic Table* Leon Gray,2013-12-15 Provides basic information on the periodic table Includes biographical information on Dmitri Mendeleev color photographs and diagrams sidebars a glossary and further reading sources **The Periodic Table of the Elements** Richard J. Puddephatt,Patrick Kevin Monaghan,1994 The Periodic Table Eric R. Scerri,2020 The Periodic Table Its Story and Its Significance traces the evolution and development of the periodic table from Mendeleev s 1869 first published table and onto the modern understanding provided by modern physics *The Periodic Table Book* Tom Jackson,2017 Profiles every element on the periodic table and describes their properties when they were discovered and how they are used in household materials

The Periodic Table: A Very Short Introduction Eric R. Scerri,2011-10-27 Presents a modern and fresh exploration of the periodic table considering the deeper implications of the arrangements of the table to atomic physics and quantum mechanics *Basher Science: The Complete Periodic Table* Adrian Dingle,Simon Basher,Dan Green,2015-01-06 Combines science and art to create a simple way for students to learn the periodic table Elements and the Periodic Table, Grades 5 - 12 Theodore S. Abbigy,2013-01-02 Aligned to Common Core State Standards Elements and the Periodic Table present the basics of the Periodic Table in an easy to understand easy to master way It contains fun activities transparency masters quizzes tests rubrics grading sheets and more From basic elements to table organization Elements and the Periodic Table is the essential handbook for middle school science **Elements and the Periodic Table, Grades 5 - 8** Abbigy,2013-01-02 Aligned to Common Core State Standards Elements and the Periodic Table present the basics of the Periodic Table in an easy to understand easy to master way It contains fun activities transparency masters quizzes tests rubrics grading sheets and more From basic elements to table organization Elements and the Periodic Table is the essential handbook for middle school science *The Periodic Table* Sharon Katz Cooper,2007 An introduction to chemistry s periodic table of elements *The Periodic Table* ,2008 Periodic Table & Periodic Properties M. Satake,2010 The book is primarily meant for undergraduate

students of chemistry General reader who is interested in chemistry of elements and their behaviour will find it equally interesting and easy to understand Periodic Table of the Elements Ekkehard Fluck, Klaus G. Heumann, 2000-01-12 What is the ionization energy of silver The melting point of tantalum Or the half life of technetium If you know all those numbers by heart you will not need this periodic table If you don't you will find more than 3000 element data on this easy to use card 100 % error free and according to the latest IUPAC recommendations **The Periodic Table** DK, 2025-09-04 This is the periodic table as you've never seen it before By presenting the elements in the order in which they were discovered The Periodic Table also tells the story of chemistry Now iconic the table was invented in the 1800s long before anyone knew what was inside an atom As more elements were found the table got bigger Some were discovered at risk to life and limb for example the hunt for the explosive element fluorine injured or even killed several scientists who came to be known as the fluorine martyrs Even now the table is still growing as scientists push the boundaries and add to the 118 elements discovered so far Some of the newest elements have been observed only as they briefly flicker into existence in the extreme conditions inside particle colliders and accelerators Gorgeous photographs show even the most familiar elements in a new light Bold clear infographics and artworks reveal aspects of their atomic structure that explain why they behave the way they do For example why is gold so malleable and diamond so incredibly hard How can bismuth levitate and why does cobalt make paint such a vivid blue If you want to unlock the secrets of matter itself open The Periodic Table

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