

## THE MARXIST VIEW OF ANCIENT PHILOSOPHY

THE subject of this paper is the application of the Marxist interpretation of history to ancient philosophy in two recent books: Mr Farrington's *Science and Politics in the Ancient World* and Mr George Thomson's *Aeschylus and Athens*. Mr Thomson's book covers a much wider field; his references to the philosophers are only incidental. But he seems to share Mr Farrington's view of them; and he is an avowed believer in the Marxist doctrine.

A word must first be said about this doctrine. I will quote one statement of it from a work which is still cited as authoritative.

Engels, in his polemic against the unfortunate Dühring,<sup>1</sup> declared that the conception of history had been decisively changed by the 'new facts' of the working-class movements in the 1830's and 1840's. 'The new facts', he says, 'made imperative a new examination of all past history, and then it was seen that all past history was the history of class struggles, that these warring classes of society are always the product of the modes of production and exchange, in a word of the *economic* conditions of their time; that therefore the economic structure of society always forms the real basis from which, in the last analysis, is to be explained the whole superstructure of legal and political institutions, as well as the religious, philosophical, and other conceptions of each historical period. Now idealism was driven from its last refuge, the philosophy of history; now a materialist conception of history was propounded, and the way found to explain man's consciousness by his being, instead of, as heretofore, his being by his consciousness.'

This is a very sweeping statement. I shall try to follow out its implications in a train of thought which can be traced in Mr Farrington's treatment of the history of philosophy, and in particular in his view of Epicurus and Plato.

<sup>1</sup> *Anti-Dühring* (English trans.), p. 32.

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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414l 22 414 l we can substitute 101 325kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k for temperature into the ideal gas equation and solve for r r

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