AIR, COMBUSTION, RUSTING AND FIRE 17.A sample of iron rusts. Explain w

- Extinguishing a fire caused by burning kerosene, carbon dioxide is used in preference to water. Explain why?
- Give a reason why, carbon dioxide is used as a fire extinguisher?
- Carbon dioxide is not used as a fire extinguisher in class D fires. Explain why?
- Burning magnesium continue to burn inside the gas jar full of carbon dioxide. Explain why?
- 5. A piece of burning magnesium continues to burn in a gas jar of Sulphur dioxide gas. Explain why?
- A small child has been playing with matches and its clothes have caught fire. Explain how you would put out the flames.
- Zinc is used to cover iron so as to prevent rusting of iron. Explain why?
- 8. Explain why motor vehicles based in Mombasa rust much faster than those based in Nairobi?
- If the tin surface is scratched then the iron will rust. Why does this not happen with zinc plating (galvanizing)?
- 10. Aluminium metal would protect iron from rusting better than zinc metal. Explain why?
- 11.It is not advisable to wash vehicles using sea water, Explain why?
- 12. Why do car exhaust rust faster than other structures made of steel?
- 13. The parts of shipwrecks made of iron rust more slowly in deep water. Explain why?
- 14.Alkaline medium inhibits the rusting of iron. Explain why?
- 15. Iron does not rust even if zinc coating is broken in a galvanized iron pipe. Explain why?
- 16.Zinc is used to coat for iron and not vice versa. Explain why?

- 17.A sample of iron weighs more after it rusts. Explain why?
 - 18. The percent of carbon dioxide in the atmosphere does not significantly vary from 0.04%. Explain why?
- 19. The air over an open flame is always moist. Explain why?
- 20.Explain why, cooking pots made up of aluminium do not corrode easily when exposed to air?
- 21. Why does a magnesium ribbon be cleaned before burning in air?
- 22. When a candle was burnt completely, the total m ass of the products was found to be greater than the original mass of the candle. Explain why?
- 23.In temperate countries, salt is sprayed on the roads to defrost and clear roads but the long-term effect on this practice is costly to motorists. Explain why?
- 24. Why baking soda does extinguish fire?
- 25. Tetrachloromethane is used as a fire extinguisher but not carbon disulphide. Explain why?
- 26.Explain why magnesium gains in mass but coal loses mass when each is burned in air?

HEAT SOURCES AND FALAMES

- when air holes of a Bunsen burner are fully opened the Bunsen-burner produce a non- luminous flame. Explain why?
- It is advisable to set the Bunsen burner to luminous flame prior to experiment. Explain why?
- it is advisable to turn a non-luminous flame to luminous flame while the flame is not in use. Explain why?
- 4. Non -luminous flame is used for heating. Explain why by giving two reasons?
- Explain how hotness of Bunsen burner flame can be increased.

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