2.) Cyclopropane is a compound of carbon and hydrogen that is used as a general anesthetic. When 1.00 g of this substance was burned completely in oxygen, 3.14 g of CO₂ and 1.29 g of H₂O were produced. What is the empirical formula of the compound? The molar mass of cyclopropane is about 42 g/mole.

0.0714:0.1433

0.1433 moles & H.

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United States. Army. Corps of Engineers

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