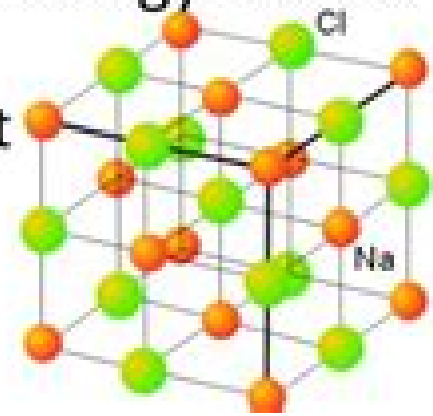


Nature of Solids

Basic Properties:

- Solids, unlike liquids, have atoms and molecules in fixed positions
 - solids have strong intermolecular forces
 - Solids have a definite shape and volume
- The density of most solids is only slightly greater than the liquid phase of the same material
 - Exception is with water
- KE of Molecules (atoms) is less than the energy it takes to separate them
 - A solid become liquid at its melting point
 - $KE = \text{intermolecular forces}$



Nature Of Solids

Hussin A.Rothana



Nature Of Solids:

Unveiling the Magic of Words: A Report on "**Nature Of Solids**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is actually awe-inspiring. Enter the realm of "**Nature Of Solids**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book's central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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