

II. PREPARATION OF ACETYL SALICYLIC ACID (ASPIRIN):

1. Complete the table below by entering appropriate values into the empty boxes. Show calculations for all entries clearly below.

Reactants	Chemical Formula ($C_xH_yO_z$)	MW (g/mol) ± 0.01 g	Grams of Reactant	Moles of Reactant	Moles of Aspirin ¹	Limiting Reagent (put X)	Melting Range
Salicylic Acid							
Acetic Anhydride							

- 1 The moles of aspirin from each reactant if that reactant were the limiting reagent (show calculations below).
- 2 Calculate the grams of acetic anhydride from the density = _____ g/mL and the volume you used: _____ mL. Show work below:

Show all calculations for the above table, below making sure to include units:

Molecular Weight: Acetic Anhydride
(use atomic weights ± 0.01 g)

Molecular Weight: Salicylic Acid
(use atomic weights ± 0.01 g)

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Table of Contents Pkg Acp Chem 104l

1. Understanding the eBook Pkg Acp Chem 104l
 - The Rise of Digital Reading Pkg Acp Chem 104l
 - Advantages of eBooks Over Traditional Books
2. Identifying Pkg Acp Chem 104l
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Pkg Acp Chem 104l
 - User-Friendly Interface
4. Exploring eBook Recommendations from Pkg Acp Chem 104l
 - Personalized Recommendations
 - Pkg Acp Chem 104l User Reviews and Ratings
 - Pkg Acp Chem 104l and Bestseller Lists
5. Accessing Pkg Acp Chem 104l Free and Paid eBooks

- Pkg Acp Chem 104l Public Domain eBooks
- Pkg Acp Chem 104l eBook Subscription Services
- Pkg Acp Chem 104l Budget-Friendly Options
- 6. Navigating Pkg Acp Chem 104l eBook Formats
 - ePub, PDF, MOBI, and More
 - Pkg Acp Chem 104l Compatibility with Devices
 - Pkg Acp Chem 104l Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Pkg Acp Chem 104l
 - Highlighting and Note-Taking Pkg Acp Chem 104l
 - Interactive Elements Pkg Acp Chem 104l
- 8. Staying Engaged with Pkg Acp Chem 104l
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Pkg Acp Chem 104l
- 9. Balancing eBooks and Physical Books Pkg Acp Chem 104l
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Pkg Acp Chem 104l
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Pkg Acp Chem 104l
 - Setting Reading Goals Pkg Acp Chem 104l
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Pkg Acp Chem 104l
 - Fact-Checking eBook Content of Pkg Acp Chem 104l
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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