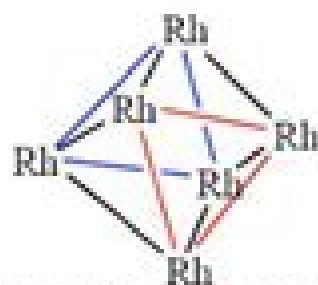


# Small metal clusters

*The simplest metal cluster contains the M<sub>3</sub> unit, held together by metal-metal bonds (rather than solely by bridging ligands). An example is Ru<sub>3</sub>(CO)<sub>12</sub>. The trigonal arrangement of metal atoms is reminiscent of the close-packing of spheres in bulk metal. The cluster Rh<sub>6</sub>(CO)<sub>16</sub> has an octahedral arrangement of metal atoms (with 12 terminal CO ligands and 4 triply-bridging CO ligands). The metal organization is analogous to the packing of two trigonal M<sub>3</sub> units in adjacent layers in bulk metal.*



structure of Ru<sub>3</sub>(CO)<sub>12</sub>



framework of Rh<sub>6</sub>(CO)<sub>16</sub>



# Metal Clusters In Chemistry

**LL Leslie**



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