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**SPSS for Macintosh Made Simple** Colin D. Gray,Paul R. Kinnear,1998 SPSS for Macintosh Made Simple is an introductory guide for the Macintosh user This book has all the features of the successful shows the reader how to explore gives advice on choosing a statistical test includes important cautions illustrates techniques with fully annotated SPSS menus dialog boxes contains an abundance of worked examples has a comprehensive table of contents index

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web understand that parts of an acid base titration be able to determine the  $k_a$  or  $k_b$  from ph data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate  $k_{a1}$  and  $k_{a2}$  for a polyprotic acid by the end of this lab students should be able to

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web 1 approximately 3 g of hydrated oxalic acid h 2 c 2 o 4 2h 2 o was weighted to the nearest 0 g in a 50 ml beaker 2 about 30 ml of distilled water was added to the beaker to dissolve the oxalic acid 3 the solution was transferred into 250 ml volumetric flask the beaker was rinsed and its contents were poured into the flask

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web solution a titrant volume 0 ml the solution ph is due to the acid ionization of hcl because this is a strong acid the ionization is complete and the hydronium ion molarity is 0 100 m the ph of the solution is then ph log 0 100 1 000 ph log 0 100 1 000 b titrant volume 12 50 ml

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web one type of titration uses a neutralization reaction in which an acid and a base react to produce a salt and water in equation 1 the acid is hcl hydrochloric acid and the base is naoh sodium hydroxide when the acid and base react they form nacl sodium chloride which is also known as table salt

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