



Red Spotted Newt

ML Morrison

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 calculation correctly the reaction of an acid with a base to make a salt and water is a common reaction in the laboratory
 partly because so many compounds can act as acids or bases

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 a 25.0 mL sample of 0.100 M hydrochloric acid with 0.100 M sodium hydroxide the values of the pH measured after successive
 additions of small amounts of NaOH are listed in the first column of this table and are graphed in

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 an acidic solution of known concentration a standard solution is added to a basic alkaline solution of unknown concentration
 until the solution is exactly neutralised i.e. there is only salt and water it is possible to calculate

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