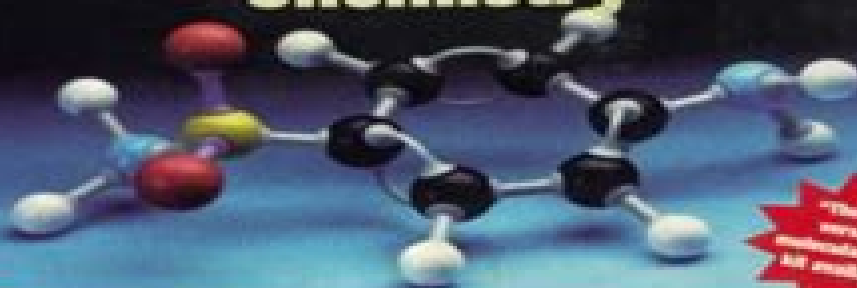


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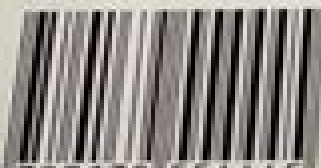
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Grundlagen der organischen Chemie Alfons Hädener,Heinz Kaufmann,2006-11-21 Die 11 Auflage dieser etablierten Einführung wurde sorgfältig bearbeitet und erweitert Das Buch vermittelt Studierenden der Chemie im Haupt und Nebenfach das Verständnis für die Grundprinzipien der organischen Chemie und ermöglicht darüber hinaus auch eine Vertiefung des in der Schule gebotenen Lehrstoffes Das bewährte Konzept die Bindungsverhältnisse in organischen Molekülen ihren räumlichen Bau und die grundlegenden Reaktionstypen getrennt von Nomenklatur und Systematik zu behandeln wurde beibehalten Metallorganische Reagenzien Methoden der asymmetrischen Synthese und die Prinzipien katalytischer Prozesse werden ebenso berücksichtigt wie die wichtigsten Methoden zur Charakterisierung organischer Verbindungen Eine Übersicht über eine Reihe von Naturstoffklassen vermittelt die Grundlagen zum Verständnis biochemischer Prinzipien Zahlreiche Übungen vertiefen den Stoff und erleichtern die Selbstkontrolle **Chemistry** John McMurry,1998 *The Prentice Hall Molecular Model Set for Organic Chemistry* ,1983 This kit enables users to build virtually all simple molecules encountered in organic chemistry Includes space filling models that simulate the true shape of saturated compounds Provides open models that form realistic single double and triple bonds even strained rings Allows smooth rotation of the bonds to make conformational analysis easy Contains enough components to create several models at once The components are precision tooled from quality plastics are virtually indestructible and come in a sturdy plastic case for easy storage Provides a useful Instruction Book with photos diagrams and concise discussions of chemical principles The Prentice Hall Molecular Model Set for Organic Chemistry John W. Lehman,1984 Resumen This kit enables users to build virtually all simple molecules encountered in organic chemistry top Features includes space filling models that simulate the true shape of saturated compounds provides open models that form realistic single double and triple bonds even strained rings allows smooth rotation of the bonds to make conformational analysis easy contains enough components to create several models at once the components are precision tooled from quality plastics are virtually indestructible and come in a sturdy plastic case for easy storage provides a useful Instruction Book with photos diagrams and concise discussions of chemical principles Set Includes 28 Hydrogen white 14 Carbon black 8 Oxygen red 4 Nitrogen blue 8 Chlorine green 2 Bromine orange 2 Iodine purple 40 Single Bond Space filling Models white 40 Single Bond Open Models short grey 12 Double and Triple Bonds Open Models long grey *Aldrich Advancing Science* ,2005 Advanced Organic Chemistry Francis A. Carey,2013-06-29 The purpose of this edition is the same as that of the first edition that is to provide a deeper understanding of the structures of organic compounds and the mechanisms of organic reactions The level is aimed at advanced undergraduates and beginning graduate students Our goal is to solidify the student's understanding of basic concepts provided in an introduction to organic chemistry and to fill in

much more information and detail including quantitative information than can be presented in the first course in organic chemistry The first three chapters consider the fundamental topics of bonding theory stereochemistry and conformation Chapter 4 discusses the techniques that are used to study and characterize reaction mechanisms The remaining chapters consider basic reaction types with a broad coverage of substituent effects and stereochemistry being provided so that each reaction can be described in good if not entirely complete detail The organization is very similar to the first edition with only a relative shift in emphasis having been made The major change is the more general application of qualitative molecular orbital theory in presenting the structural basis of substituent and stereoelectronic effects The primary research literature now uses molecular orbital approaches very widely while resonance theory serves as the primary tool for explanation of structural and substituent effects at the introductory level Our intention is to illustrate the use of both types of interpretation with the goal of facilitating the student's ability to understand and apply the molecular orbital concepts now widely in use

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Rogers, 2006-09-06 Heats of hydrogenation constitute a body of thermochemical information that has had an on going significance despite the small number of research groups engaged in the work Recent highly accurate quantum mechanical calculations requiring reference standards of high accuracy have brought hydrogen thermochemistry back into contemporary focus This book concentrates on distinctive features of hydrogen thermochemistry such as the practical and historical aspects of experimental determination of the enthalpies of hydrogenation and formation of organic compounds primarily hydrocarbons literature on hydrogen thermochemistry over the last 70 years as well as the impact of contemporary advances in computer hardware and software on the calculation of heats of hydrogenation

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