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# Lotus Notes And Domino 5 Bible

**Joseph Lowery, Derren Whiteman**



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web the balanced chemical equation for this reaction is  $\text{Mg}(\text{s}) + 2\text{HNO}_3(\text{aq}) \rightarrow \text{Mg}(\text{NO}_3)_2(\text{aq}) + \text{H}_2(\text{g})$  in this equation the s symbolizes solid magnesium aq represents aqueous nitric acid aq signifies magnesium nitrate in a dissolved state and g represents the gaseous hydrogen produced

[\$\text{HNO}\_3 + \text{C}\_2\text{H}\_3\text{O}\_2 \rightarrow \text{HC}\_2\text{H}\_3\text{O}\_2 + \text{Mg}\(\text{NO}\_3\)\_2\$  chemical equation](#) - Aug 11 2023

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web magnesium hydroxide nitric acid balanced total equation  $\text{Mg(OH)}_2(\text{aq}) + 2\text{HNO}_3(\text{aq}) \rightarrow \text{Mg(NO}_3)_2(\text{aq}) + 2\text{H}_2\text{O}$  total ionic equation  $\text{Mg}^{2+}(\text{aq}) + 2\text{OH}^{-}(\text{aq}) + 2\text{H}^{+}(\text{aq}) + 2\text{NO}_3^{-}(\text{aq}) \rightarrow \text{Mg}^{2+}(\text{aq}) + 2\text{NO}_3^{-}(\text{aq}) + 2\text{H}_2\text{O}$  net ionic equation  $2\text{OH}^{-}(\text{aq}) + 2\text{H}^{+}(\text{aq}) \rightarrow 2\text{H}_2\text{O}$  sulfuric acid and lithium hydroxide balanced total equation  $\text{H}_2\text{SO}_4(\text{aq}) + 2\text{LiOH}(\text{aq}) \rightarrow \text{Li}_2\text{SO}_4(\text{aq}) + 2\text{H}_2\text{O}$