

INTRODUCTION TO ELECTRICITY

- **Electricity** - electron in motion
- **William Gilbert** - Father of electricity and English physicist
 - Published:**
 - **Electric Attraction**
 - **Electric Force**
- **Electron**- negative charge of electricity
- **Proton** - positive charge of electricity

Theories

- All matter is made up of molecules.
- Molecules are made up of atoms.
- Atom contains neutrons, electrons and protons
- Neutron is neutral. It is neither positively nor negatively charged.
- The electron of an atom of any substance could be transformed into another atom.

Factors that Influence Conductor Resistance

- **Composition of the Conductor** - This refers to a conductor having free electrons that have low resistance.
- **Length of Wire** - The longer the wire, the higher resistance.
- **Cross-Sectional Area of Wire** - The bigger the cross-sectional area of the wire, the lower its resistance.
- **Temperature** - Metal offers high resistance to high temperature (heat).

Electric Current

is the flow or rate of flow of electric force in a conductor. A current will only flow if a circuit is formed comprising a

complete loop and contains all the following required components.

1. Source of voltage
2. A closed loop of wiring
3. An electric load
4. A means of opening and closing the circuit

ELECTRIC CURRENT IS CLASSIFIED AS:

- **Direct Current (DC)** - flows in one direction. The flow is said to be from negative to positive.
- **Alternating Current (AC)** - electricity constantly reverses its direction of flow. It is generated by a machine called a generator.

Comparison of AC and DC Electricity

- **DC** - Power is the product of voltage and current

$$\text{Watts} = \text{Volts} \times \text{Amperes}$$

- **AC** - the product of volts and amperes is equal to the quantity called volt-ampere

$$\text{Volt Amperes} = \text{Volts} \times \text{Amperes}$$

- Conversion of volt-ampere to watts or power

$$W = V \times I \times pf$$

- However, both are expressed in the unit of measure called **Ohms**.
- **Impedance** is the electrical term for Friction in AC electricity.
- In a direct current DC circuit, this impedance is called **Resistance**

Practical Dictionary Of Electricity Electron

P. H. Sydenham



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A Practical Medical Dictionary ... Thomas Lathrop Stedman,1922 *The Practical Dictionary of Electricity and Electronics* Ruth L. Oldfield,1959 **Practical Course in Terminology Processing** Juan C. Sager,1990-01-01 Since the advent of the computer terminology management can be carried out by almost anyone who has learnt to use a computer Terminology management has proved to be an efficient tool in international communications in industry education and international organisations Software packages are readily available and international corporations often have their own terminology database Following these developments translators and terminologists are confronted with a specialised form of information management involving compilation and standardisation of vocabulary storage retrieval and updating A Practical Course in Terminology Processing provides the key to methods of terminology management for the English language for general and specific purposes This unique course has been developed on the basis of years of teaching experience and research at the University of Manchester Institute of Science and Technology UMIST UK and is particularly suitable for translation courses freelance translators technical writers as well as for non linguists who are confronted with terminology processing as part of their profession The 1996 reprint of the paperback edition includes an index *Practical Microwave Electron Devices* Bozzano G Luisa,2013-10-22 Practical Microwave Electron Devices provides an understanding of microwave electron devices and their applications All areas of microwave electron devices are covered These include microwave solid state devices including popular microwave transistors and both passive and active diodes quantum electron devices thermionic devices including relativistic thermionic devices and ferrimagnetic electron devices The design of each of these devices is discussed as well as their applications including oscillation amplification switching modulation demodulation and parametric interactions Numerous design examples and case studies are presented throughout the book For each microwave electron device covered typical design examples or case studies are presented as well as qualitative or quantitative explanations The fundamental theory of each device is summarized along with the underlying principles of the design Each summary is presented so that the design techniques can be applied to other specific cases designs and applications Review questions are included with each chapter to stimulate creative thinking and enhance the acquisition of knowledge and design skills This book is written for engineers scientists and technicians seeking practical knowledge on microwave electron devices and their applications through self study It is also suitable for use as a college textbook in upper division courses for seniors and first year graduate students in electrical engineering *NBS Special Publication* ,1964 **Foreign-language and English Dictionaries in the Physical Sciences and Engineering** Tibor W. Marton,1964 The bibliography lists over 2800 unilingual bilingual and polyglot dictionaries glossaries and encyclopedias in the physical sciences engineering and technology published during the past twelve years The majority of the titles cited have English as the source or target language or are dictionaries giving definitions in English The bibliographic entries are arranged in 49 subject classes within

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web other articles where vapour pressure curve is discussed thermodynamics the clausius clapeyron equation pressure remains equal to the vapour pressure p_{vap} as the piston moves up as long as both phases remain present all that happens is that more water turns to steam and the heat reservoir must supply the latent heat of vaporization λ 40 65

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web mar 13 2023 the measured values τ_{ign} is represented by the data for n decane t 1600 k at p 3 5 atm n decane t 1175 k p 5 atm n decane t 1300 k p 13 atm n decane and kerosene jet a t 1750 k p 9 atm jet a kerosene t 1320 k p 11 atm and jet a kerosene t 1380 k p 10 atm

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web a the vapor pressure curve of water intersects the p 1000 mmhg line at about 110 c this is therefore the boiling point of water at 1000 mmhg b the vertical line corresponding to 250 c intersects the vapor pressure curve of mercury at p 75 mmhg hence this is the pressure required for mercury to boil at 250 c

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web an examination of methods for calculating vapor pressure of petroleum hydrocarbon alan e zengel 1964 at high temperatures the vapor pressure of kerosene type hydrocarbons is difficult to measure accurately with laboratory apparatus be cause of difficulties in direct measurement of vapor pressure temperature

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web jan 4 2021 vapor pressure 5 mmhg at 100 f flash point 100 162 f vapor density 4 5 specific gravity 0 81 ionization

potential lower explosive limit lel 0.7 upper explosive limit uel 5 nfpa health rating

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web vapour pressure of kerosene hydrocarbon the vapor pressure of a liquid is defined as the pressure exerted by the molecules that escape from the liquid to form a separate vapor phase above the liquid surface this pressure is formed in a thermodynamic equilibrium state in a closed container at a certain temperature

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web allow the gaseous phase thermodynamics to be curve fitted to 300k about 142k lower than the normal boiling point 542k 2 of kerosene nevertheless in actual calculations involving liquid kerosene fuel the heat capacity and latent heat of vaporization of liquid kerosene 12 are used for

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web as a general trend vapor pressures of liquids at ambient temperatures increase with decreasing boiling points this is illustrated in the vapor pressure chart see right that shows graphs of the vapor pressures versus temperatures for a variety of liquids

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web the vapor pressure of a liquid is defined as the pressure exerted by the molecules that escapes from the liquid to form a separate vapor phase above the liquid surface the pressure exerted by the vapor phase is called the vapor or saturation pressure vapor or saturation pressure depends on temperature

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web sep 21 2022 a vapor pressure curve is a graph of vapor pressure as a function of temperature to find the normal boiling point of liquid a horizontal line is drawn from the y axis at a pressure equal to standard pressure

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web the graph of the vapor pressure of water versus temperature in figure page index 3 indicates that the vapor pressure of water is 68 kpa at about 90 c thus at about 90 c the vapor pressure of water will equal the atmospheric pressure in

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web nov 1 2003 droplet evaporation modeling by the distillation curve model accounting for kerosene fuel and elevated pressures

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web vapor pressure 0.1 psi at 100 f ntp 1992 vapor density relative to air 4.5 ntp 1992 heavier than air will sink specific gravity 0.8 at 59 f uscg 1999 less dense than water will float

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web in fact when the temperature increases from 20 to 60 c the vapor pressure of kerosene rises from 4 to 18 mbar 34 this suggests that kerosene cannot be entirely vaporized at 60 c

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web kerosene icsc 0663 physical chemical information physical state appearance low viscosity liquid with characteristic odour physical dangers as a result of flow agitation etc electrostatic charges can be generated vapour pressure kpa at 37.8 c 13.7 exposure health effects