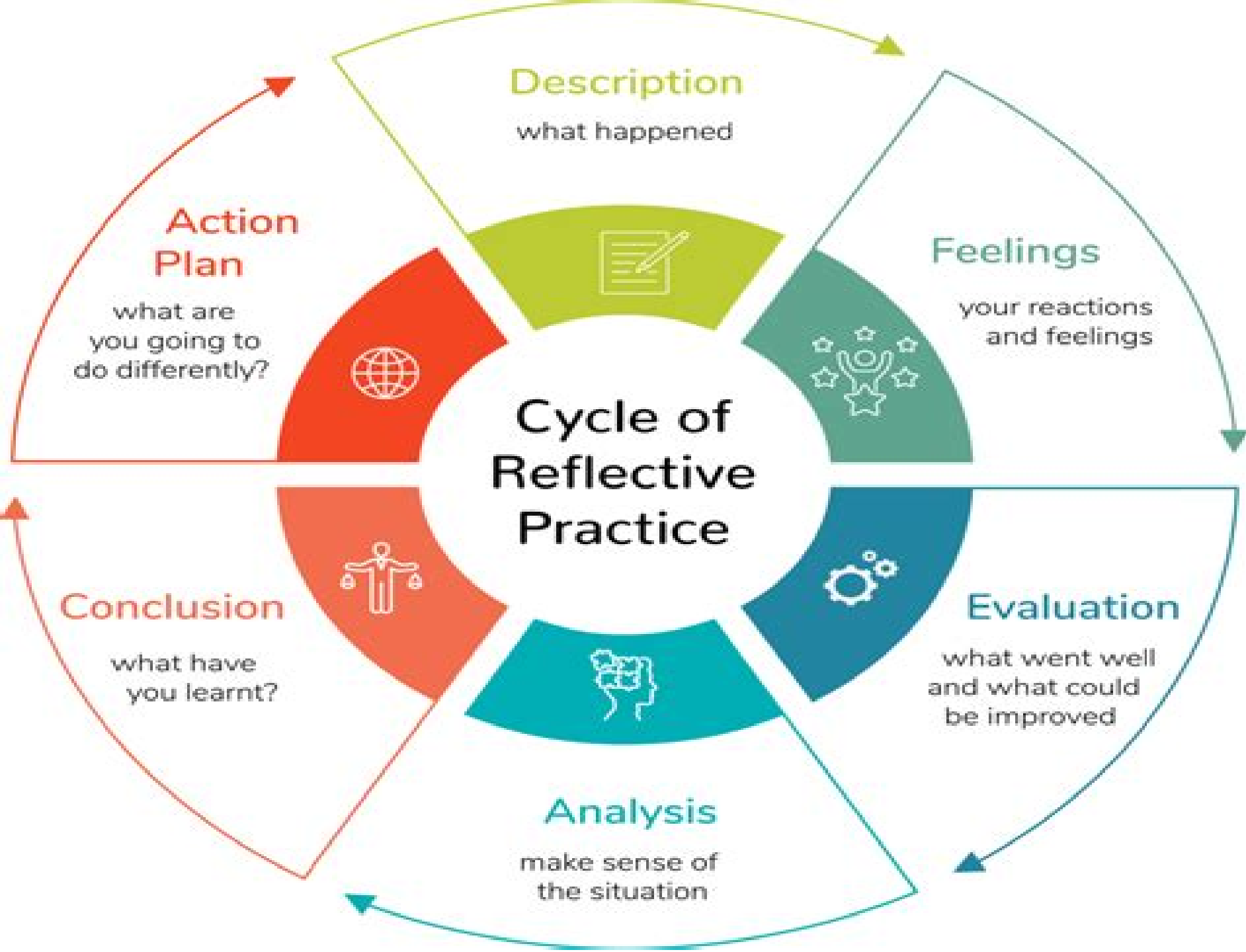




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Mark D. Groza, Charles B. Ragland



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web background this experiment in designed to study the kinetics of a chemical reaction the reaction involves the oxidation of iodide ions by bromate ions in the presence of acid $6\text{I}^- \text{aq} + \text{BrO}_3^- \text{aq} + 6\text{H}^+ \text{aq} \rightarrow 3\text{I}_2 \text{aq} + \text{Br}^- \text{aq} + 3\text{H}_2\text{O} \text{l}$ reaction 1 the reaction is somewhat slow at room temperature the reaction rate depends

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