

CHEM 232 : GENERAL CHEMISTRY III

Transcript title

General Chemistry III

Credits

3

Grading mode

Letter grade

Total contact hours

48

Lecture hours

48

Lab hours

0

Prerequisites

None

Course Description

Advanced topics in general chemistry including equilibrium, electrochemistry, thermodynamics, and kinetics.

Course objectives: students will be able to: understand the principles of thermodynamics, electrochemistry, and kinetics; apply these principles to solve problems; and understand the relationship between thermodynamics, electrochemistry, and kinetics.

Course learning outcomes

1. Calculate the equilibrium constant for a reaction.
2. Calculate the standard Gibbs free energy of formation for a compound.
3. Calculate the standard Gibbs free energy of formation for a compound.
4. Calculate the standard Gibbs free energy of formation for a compound.
5. Calculate the standard Gibbs free energy of formation for a compound.
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understand the relationship between thermodynamics, electrochemistry, and kinetics.

1. Calculate the equilibrium constant for a reaction.

2. Calculate the standard Gibbs free energy of formation for a compound.

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