



## GEOLOGY

# Seafloor hydrothermal systems control long-term changes in seawater $[\text{Li}^+]$ : Evidence from fluid inclusions

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Secular variations in the major ion chemistry and isotopic composition of seawater on multimillion-year time scales are well documented, but the causes of these changes are debated. Fluid inclusions in marine halite indicate that the Li concentration in seawater  $[\text{Li}^+]_{\text{sw}}$  declined sevenfold over the past 150 million years (Ma) from  $\sim 184 \mu\text{mol/kg H}_2\text{O}$  at 150 Ma ago to  $27 \mu\text{mol/kg H}_2\text{O}$  today. Modeling of the lithium geochemical cycle shows that the decrease in  $[\text{Li}^+]_{\text{sw}}$  was controlled chiefly by long-term decreases in ocean crust production rates and mid-ocean ridge and ridge flank hydrothermal fluxes without requiring changes in continental weathering fluxes. The decrease in  $[\text{Li}^+]_{\text{sw}}$  parallels the 150 Ma increase in seawater  $\text{Mg}^{2+}/\text{Ca}^{2+}$  and  $^{87}\text{Sr}/^{86}\text{Sr}$ , and the change from calcite to aragonite seas, KCl to  $\text{MgSO}_4$  evaporites, and greenhouse to icehouse climates, all of which point to the importance of plate tectonic activity in regulating the composition of Earth's hydrosphere and atmosphere.

## INTRODUCTION

The major ion ( $\text{Mg}^{2+}$ ,  $\text{Ca}^{2+}$ ,  $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{SO}_4^{2-}$ ,  $\text{Cl}^-$ ,  $\text{HCO}_3^-$ ) (1) and isotopic composition (i.e.,  $\delta^7\text{Li}$ ,  $^{87}\text{Sr}/^{86}\text{Sr}$ ) (2, 3) of seawater have varied in the Phanerozoic Eon. Systematic changes in the  $\text{Mg}^{2+}$ ,  $\text{Ca}^{2+}$ , and  $\text{SO}_4^{2-}$  concentrations in Phanerozoic seawater regulate the alternating  $\text{MgSO}_4$  and KCl mineralogies of marine potash evaporites (4). Variations in the  $\text{Mg}^{2+}$  and  $\text{Ca}^{2+}$  concentrations may have influenced the evolution of marine shell-building organisms, including dominant reef-builders and skeletal algae, because the  $\text{Mg}^{2+}/\text{Ca}^{2+}$  ratio of seawater controls which carbonate, calcite, or aragonite precipitates during so-called calcite seas and aragonite seas (5). The  $\text{Mg}^{2+}/\text{Ca}^{2+}$  ratio of seawater from the Jurassic to the present closely matches the marine  $^{87}\text{Sr}/^{86}\text{Sr}$  curve, and both show a sharp rise in the Cenozoic similar to that exhibited by the marine lithium isotope ( $\delta^7\text{Li}$ ) record from 60 to 0 million years (Ma) ago (Fig. 1) (3, 6, 7). Long-term changes in the major ion chemistry of seawater broadly correlate with variations in atmospheric  $\text{CO}_2$  and icehouse-greenhouse climate, suggesting links to the carbon cycle (8).

Long-term variations in seawater composition are now well documented, but there is disagreement over which input or output fluxes were responsible for those changes. Proposed controls of temporal changes in seawater chemistry include variations in the rates of (i) ocean crust production and mid-ocean ridge (MOR) hydrothermal discharge (4), (ii) river flux (RFV) and terrestrial weathering intensity (9), (iii) deposition of marine carbonates and dolomitization (10), (iv) low-temperature, off-axis alteration of oceanic crust (AOC) (11), and (v) formation of marine authigenic aluminosilicate clays (MAAC) (12). AOC and MAAC together can be considered reverse weathering (3).

Recent advances in laser ablation inductively coupled plasma mass spectrometry (LA-ICP-MS) now allow quantitative

measurement of minor and trace elements in fluid inclusions in halite, including lithium ( $\text{Li}^+$ ) (13).  $\text{Li}^+$ , a trace element in seawater, is particularly useful for understanding the drivers of long-term changes in seawater chemistry. It is a conservative element not incorporated to any significant degree in minerals precipitated from evaporating seawater, such as calcite, aragonite, gypsum, or halite (fig. S1).  $\text{Li}^+$  is entirely derived from terrestrial silicate mineral weathering and seafloor-basalt interactions with hydrothermal fluids at MORs, and its concentration in seawater is not significantly influenced by biological processes or precipitation of carbonates (3).  $\text{Li}^+$  today is supplied to the oceans by MOR hydrothermal fluids ( $F_{\text{MOR}} = 13.7 \pm 2.8 \times 10^6 \text{ mol/year}$ ,  $58 \pm 12\%$  of global input flux) and rivers ( $F_{\text{RFV}} = 9.9 \times 10^6 \text{ mol/year}$ ,  $\sim 42\%$ ) (14, 15).  $[\text{Li}^+]$  in MOR hydrothermal fluids are three orders of magnitude higher than rivers ( $457 \pm 66$  versus  $0.265 \mu\text{mol/kg}$  of  $\text{H}_2\text{O}$ , here termed  $\mu\text{molal}$ ) and  $\sim 17 \pm 3$  times that of seawater ( $27 \mu\text{molal}$ ) (15–17). Small fluctuations in ocean crust production rates and hydrothermal fluxes could therefore change seawater  $\text{Li}^+$  ( $[\text{Li}^+]_{\text{sw}}$ ).  $\text{Li}^+$  is removed from seawater during low-temperature (off-axis) AOC ( $F_{\text{AOC}} = 2$  to  $21 \times 10^6 \text{ mol/year}$ ) (18, 19) and MAAC formation ( $F_{\text{MAAC}} = 11$  to  $20 \times 10^6 \text{ mol/year}$ ) (3, 20).

Recent records of  $\delta^7\text{Li}_{\text{sw}}$  in Cenozoic seawater from foraminifera (3) and brachiopods (21) show an increase of  $\sim 8$  to  $9\%$  over the past 60 Ma, which is thought to be due to (i) increased tectonic uplift that increased continental clay mineral formation (3), (ii) increased diagenetic reactions of marine clays (20), (iii) increased land surface reactivity and erosion (22), (iv) increased Li isotopic fractionation between seawater and oceanic crust (18), and (v) decreased continental soil formation (23). Modeling and interpretation of the Cenozoic  $\delta^7\text{Li}_{\text{sw}}$  record, however, are based on the assumption that the  $[\text{Li}^+]_{\text{sw}}$  has been roughly constant (3, 18, 22, 23). Li and West (20) tested 12 model scenarios to explain the Cenozoic  $\delta^7\text{Li}_{\text{sw}}$  record and suggested that knowledge of  $[\text{Li}^+]$  in ancient seawater would help distinguish the major controls responsible for changes in the global Li cycle. Here, we show from measurements of fluid inclusions in marine halite that  $[\text{Li}^+]_{\text{sw}}$  in Jurassic and Cretaceous seawater was up to seven times greater than in modern seawater. Such variations are not included in any

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# Seafloor Spreading Centers Hydrothermal Systems

## Benchmark Papers In Geology Number 56

**Michael Rajnik**



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