

Noble Gas Compounds

Table 22.1 Properties of Xenon Compounds

Compound	Oxidation State of Xe	Melting Point (°C)	ΔH_f° (kJ/mol) ^a
XeF ₂	+2	129	-109(g)
XeF ₄	+4	117	-218(g)
XeF ₆	+6	49	-298(g)
XeOF ₄	+6	-41 to -28	+146(l)
XeO ₃	+6	— ^b	+402(s)
XeO ₃ F ₂	+6	31	+145(s)
XeO ₄	+8	— ^c	—

^aAt 25 °C, for the compound in the state indicated.

^bA solid; decomposes at 40 °C.

^cA solid; decomposes at -40 °C.

- Discovered in 1962
- Xe forms most of these compounds
- Lower ionization energy and expanded valence shell lead to reactions.
- KrF₂ (decomposes at -10 °C) and a few other Kr compounds (unstable above -40 °C) also can be made.

Noble Gas Compounds

A. K. Tyagi, Raghumani S. Ningthoujam

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The Chemistry of the Noble Gases Cedric L. Chernick, 1967 **Noble Gases and Their Compounds** G. J. Moody, John David Ronald Thomas, 1964 Noble-gas Compounds Herbert Hyman Hyman, 1963 Noble-gas Chemistry John H. Holloway, 1968 **Noble gases**, 1970 *The Noble Gases* Adam Furgang, 2010-01-15 This book explains the characteristics of the noble gases helium neon argon krypton xenon and radon including where they are found and how they are used by humans Noble Gas Compounds a Bibliography 1962-1976 Warren E. Falconer, Neil Bartlett, 1978-02 *The Noble Gases* Howard H. Claassen, 1966 **Neon and the Noble Gases** Nigel Saunders, 2003 This series uses a common or well known element to look at the groups of the periodic table and to show the similarities and differences between elements It uses full colour illustration of the periodic table and shows the chemical symbol for each element in place alongside its neighbours Chemical formulae for common compounds are also shown Information boxes and tables contain listings of facts and figures Chemical reactions are interpreted as word equations and timelines chart the history and discovery of the elements atoms and why they don't even react with themselves It guides you through the discovery and isolation of each gas and recent experiments by scientists to make noble gas compounds It also explains why helium balloons float neon lights glow and argon atmospheres protect metals from air **Noble Gas Chemistry** Felice Grandinetti, 2018-06-27 Authored by one of the world's leading experts in the chemistry of lighter noble gases this comprehensive monograph fills the need for an up to date review of the diverse experimental techniques and theoretical methods currently in practice After reviewing the experiments breaking the paradigm of non reactive noble gases the physico chemical background is introduced Besides the emphasis on gas phase reactions the author presents other relevant systems such as chemistry in the bulk phase under high pressure and cold matrices The discussion of gas phase chemistry of the noble gases covers neutral and ionic compounds diatomic molecules complexes with small molecules and metal compounds up to large clusters *Noble-Gas Compounds. [Papers Read at the 1963 Conference on Noble-Gas Compounds Held at Argonne National Laboratory.] Edited with Introductions by Herbert H. Hyman. [With Plates.]* Argonne National Laboratory. Conference on Noble-Gas Compounds, Herbert Hyman HYMAN, 1963 **Noble Gas Compounds** Jeffery E. Fisher, 1966 **The Chemistry of the Monatomic Gases** A. H. Cockett, K. C. Smith, Neil Bartlett, 2013-10-22 The Chemistry of the Monatomic Gases presents Chapters 5 and 6 from the book *Comprehensive Inorganic Chemistry* The book deals with the monatomic gases of Group 0 of the Periodic Table The discovery origin and occurrence in nature both terrestrially and universally of monatomic gases are discussed The text also provides the group's properties highlighting their similarities and progressive change of properties with atomic weight Chemists and students studying chemistry will find the book a good reference material **The Formation of Noble Gas Compounds** Leland Cullen Allen, 1965 Halogens and Noble Gases Monica Halka, Brian Nordstrom, 2010 In spite of their adjacency in the periodic table halogens and nonmetals have very different properties

Halogens are among the most chemically reactive elements in the periodic table exhibiting a diverse chemistry in terms of the large numbers of compounds they can form. On the other hand noble gases are the least chemically reactive elements. In fact before the 1960s chemists referred to these elements as inert gases because it was believed that they exhibited no chemistry whatsoever. Providing the basics of these elements including their role in history and some of the important scientists involved in their discovery, this new full color resource features up to date scientific understanding in a clear and accessible format. Halogens and Noble Gases examines the ways humans use halogens and noble gases and the resulting benefits and challenges to society, health and the environment. Fluorine, chlorine, bromine, iodine, helium and krypton are covered in this book along with the fundamentals of chemistry and physics as well as possible future developments in halogen and noble gas science and its applications.

Elements of the p-Block C J Harding, D A Johnson, Rob Janes, 2007-10-31. Presenting a systematic approach to the chemistry of the p Block elements and hydrogen, this book also introduces some basic topics concerning chemical bonding such as oxidation numbers, bond strengths, dipole moments and intermolecular forces. The chemistry is illustrated by coverage of the biological role of nitric oxide and of hydrogen bonding and the new chemistry of carbon nanotubes. Applied aspects of the topic are developed in the two Case Studies which examine the causes and prevention of acid rain and the inorganic chemical industry. The accompanying CD ROMs cover silicate mineral structures, the inert pair effect and a database of chemical reactions of the p Block elements. The Molecular World series provides an integrated introduction to all branches of chemistry for both students wishing to specialise and those wishing to gain a broad understanding of chemistry and its relevance to the everyday world and to other areas of science. The books with their Case Studies and accompanying multi media interactive CD ROMs will also provide valuable resource material for teachers and lecturers. The CD ROMs are designed for use on a PC running Windows 95, 98, ME or 2000.

Handbook on Synthesis Strategies for Advanced Materials A. K. Tyagi, Raghuramani S. Ningthoujam, 2021-10-22. This book presents state of the art coverage of synthesis of advanced functional materials. Unconventional synthetic routes play an important role in the synthesis of advanced materials as many new materials are metastable and cannot be synthesized by conventional methods. This book presents various synthesis methods such as conventional solid state method, combustion method, a range of soft chemical methods, template synthesis, molecular precursor method, microwave synthesis, sonochemical method and high pressure synthesis. It provides a comprehensive overview of synthesis methods and covers a variety of materials including ceramics, films, glass, carbon based and metallic materials. Many techniques for processing and surface functionalization are also discussed. Several engineering aspects of materials synthesis are also included. The contents of this book are useful for researchers and professionals working in the areas of materials and chemistry.

The Formation of Noble Gas Compounds Leland C. Allen, 1965 Understanding the Atom U.S. Atomic Energy Commission, 1963 **Nuclear Science Abstracts**, 1965

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