

PERIODIC TABLE

Atomic Properties of the Elements

NIST
National Institute of
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U.S. Department of Commerce

Frequently used fundamental physical constants																		Physics Laboratory		Standard Reference Data			
For the most accurate values of these and other constants, visit physics.nist.gov/constants																		physics.nist.gov		www.nist.gov/srd			
1 second = 9 192 631 770 periods of radiation corresponding to the transition between the two hyperfine levels of the ground state of ¹³³ Cs																							
speed of light in vacuum c = 299 792 458 m s ⁻¹ (exact)																							
Planck constant h = 6.626 070 15 × 10 ⁻³⁴ J s (exact)																							
elementary charge e = 1.602 176 634 × 10 ⁻¹⁹ C																							
electron mass m _e = 9.109 383 56 × 10 ⁻³¹ kg																							
proton mass m _p = 1.672 621 9 × 10 ⁻²⁷ kg																							
fine-structure constant α = 1/137.036																							
Rydberg constant R _∞ = 10 973 731.568 162 m ⁻¹																							
Bohr magneton μ _B = 9.274 009 474 × 10 ⁻²⁴ J T ⁻¹																							
Boltzmann constant k = 1.380 658 × 10 ⁻²³ J K ⁻¹																							

Atomic Number

Ground-state Level

Symbol

Name

Atomic Weight

Ground-state Configuration

Ionization Energy (eV)

58

Ce

Cerium

140.116

[Xe]4f¹5d¹6s²

5.5387

¹Based upon ¹²C. () indicates the mass number of the longest-lived isotope.

For a description of the data, visit physics.nist.gov/data

NIST SP 966 (September 2010)

Science Of Chemistry Periodic Properties

M Carnoy



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