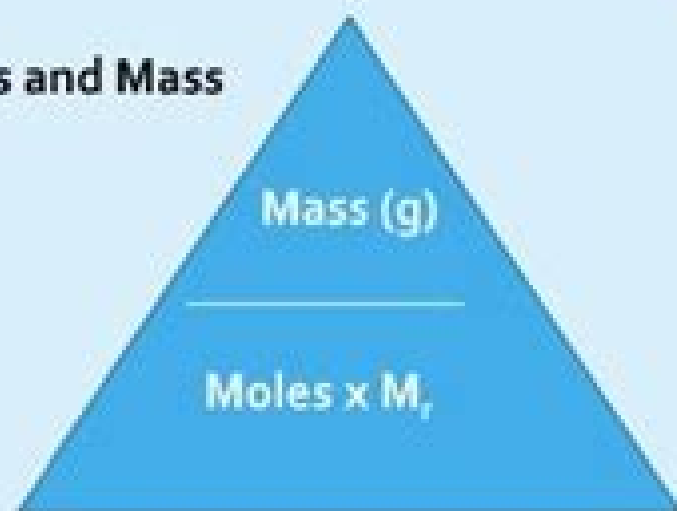




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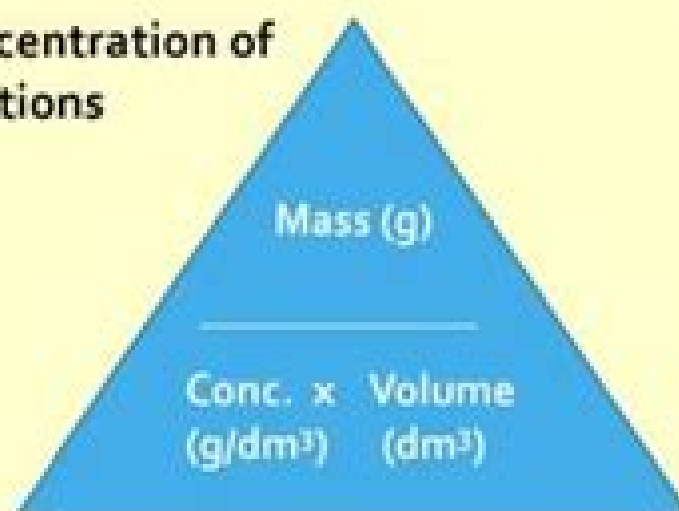
Topic 3 – Quantitative Chemistry

Moles and Mass



M_r (relative formula mass) = sum of relative atomic mass in quantities given in the formula e.g. Cl₂ = 35.5x2

Concentration of solutions



Volumes gases

$$\% \text{ Yield} = \frac{\text{Amount of product produced (g)}}{\text{Max. amount of product possible (g)}} \times 100\%$$

$$\text{Atom economy} = \frac{\text{M}_r \text{ desired product from the reaction}}{\text{Sum of M}_r \text{ of all reactants}} \times 100\%$$

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