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Systeme-D 4.0 Frank Dominguez, James S. Noblitt, 2004-06 The SYSTEME D WRITING ASSISTANT Software program provides learners with rapid access to language reference materials      **Paso a Paso 2** Addison Wesley Longman, 1997

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web rate laws and reaction order the relation between the rate of a reaction and the concentrations of reactants is expressed by its rate law for example the rate of the gas phase decomposition of dinitrogen pentoxide  $2\text{N}_2\text{O}_5 \rightarrow 4\text{NO}_2 + \text{O}_2$  has been found to be directly proportional to the concentration of  $\text{N}_2\text{O}_5$  text rate  $k$   $n$

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web the initial rate law is then measured for each of the reactions consider the reaction between nitrogen monoxide gas and hydrogen gas to form nitrogen gas and water vapor the following data were collected for this reaction at 1280 text  $\text{O}$  text  $\text{C}$  see table below notice that the starting concentrations of  $\text{CeNO}$  and  $\text{CeH}$

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web determination of a rate law and activation energy in this experiment glass beakers pipettes and a hotplate were utilized to set up the procedures a solution of deionized water a buffer potassium iodide  $\text{KI}$  sodium thiosulfate  $\text{Na}_2\text{S}_2\text{O}_3$  starch and hydrogen peroxide  $\text{H}_2\text{O}_2$  were used in order to calculate the rate of reaction  
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