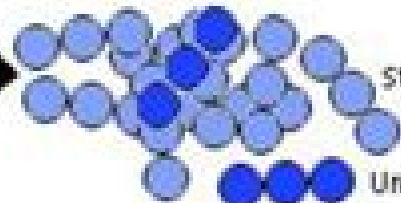
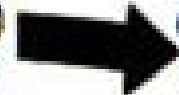


Radiometric Dating

Radiocarbon decays at a known rate. Paleontologists are able to determine the age of a fossil by measuring the amount of C-14 it contains.

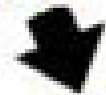


A small piece of the fossil is burned and converted to carbon dioxide gas



Stable C-12

Unstable C-14



C-14 decays into N-14, emitting an electron



Nitrogen



Electron



Living organisms absorb C-14 (radiocarbon) during their lifetimes



A radiation counter records the number of electrons emitted



Radiometric Dating Paleontologic Zonations

N Colangelo



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Bandy,1970 *Radiometric Methods of Dating Fossils* Kimberly Wylie,2006-07-16 Seminar paper from the year 2004 in the subject Archaeology University of Phoenix language English abstract Today most of the methods utilized for chronometric dating of fossils are radiometric Radiometric dating in general refers to the dating of material by using the known rate at which certain radioactive isotopes decay or at what rate there are collective changes due to radioactivity Even though isotopes of an element can be different when it comes to atomic mass the atomic number of the isotope is always the same Radioactive elements decay at unique rates dependant on the isotope This rate of decay is known as half lives it is the time necessary for 1 2 of the atoms to decay in a particular element The decay follows a geometric scale in that in the first half life of an element 1 2 of the atoms decay yet in the second half life 1 2 of those remaining decay meaning a 1 4 of the original atoms decay and so forth By measuring this decay and knowing the half life of an element scientists can date a sample

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on the basis of the lithostratigraphic and biostratigraphic principles developed during the last century. With the discovery of radioactivity the basis for a new geoscientific discipline, geochronology, was established (Rutherford, 1906). It is the study of geological time based mainly on the time signatures provided by the isotopic composition in geologic materials. The isotopic signature in a rock yields more information than that provided by the geochemical signature alone because it reflects the origin and history of the element in the rock. The aim of geochronology is to calibrate and standardize chronostratigraphic scales to develop geological time scales that have a sensitive or at least useful resolution in order to place the geological events in the correct chronological order and to assign their proper time spans. In practice the application of geochronology is much wider because the data in the natural archives often provide information on the origin, genesis and history of the materials. This of course requires an understanding of the geochemical behavior of the substances involved.

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