

Canada balsam is then added, first round the blood and then to fill up the space. The layer of blood must not be too thick. Canada balsam which is yellow and not quite clear is best suited for the purpose. The blood must remain for a few days exposed to the air, that is until crystallisation has ceased and the odour of the balsam vanished. The excess of balsam is then removed with the help of a knife, wetted with ether, turpentine, or oil of cloves; the preparation is then covered with a glass, which is fixed on with asphalt or balsam. A. K. M.

Study of Metahæmoglobin. By A. JÄDERHOLM (*Zeit. f. Biol.*, 20, 419—448).—For the purpose of obtaining metahæmoglobin crystals, the author treated dog's blood, with slight modifications, according to the sixth process described in Preyer's "Die Blutkrystalle." The crystals differ only in size and shape from those Hammarsten obtained by treating horse's blood with ferricyanide of potassium and subsequent dilution and application of cold (*Zeit. Physiol. Chem.*, 8, 186), as both forms of crystals and their solutions give the characteristic spectra of metahæmoglobin.

The addition of a very small quantity of sodium carbonate (0.00053 per cent.) to a solution, produces the so-called alkaline metahæmoglobin spectrum $\pi + \alpha_1 + \beta_1$, π being always weaker than $\alpha_1 + \beta_1$.

The bands II and III of metahæmoglobin, and $\alpha_1 + \beta_1$, of its alkaline solution, correspond very nearly with $\alpha + \beta$ of oxyhæmoglobin. The spectrum $\pi + \alpha_1 + \beta_1$ of alkaline metahæmoglobin can be obtained also by passing pure hydrogen through a solution.

If excess of sodium carbonate be avoided on the one hand, and of hydrogen on the other, the bands I and IV of metahæmoglobin are produced on shaking the solutions with air.

The latter half of the paper is devoted to the discussion of its constitution. The author explains his reason for assuming it to be a peroxidised oxyhæmoglobin, as he did in his former paper, but now agrees with Hüfner and Hülz in considering it to contain the same amount of oxygen as oxyhæmoglobin. J. P. L.

Physiological Chemistry.

Influence of Variations in the Percentage Composition of the Air on the Intensity of Respiratory Changes. By L. FRÉDÉRICQ (*Compt. rend.*, 99, 1124—1125).—The author has investigated the influence of variations in the amount of carbonic anhydride or of oxygen in the air on the intensity of respiratory changes, as measured by the amount of oxygen absorbed in the case of rabbits and of himself.

An increase in the proportion of oxygen has no effect on the absorption of oxygen in the process of respiration. When an animal is

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