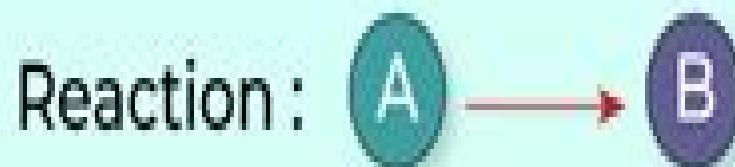
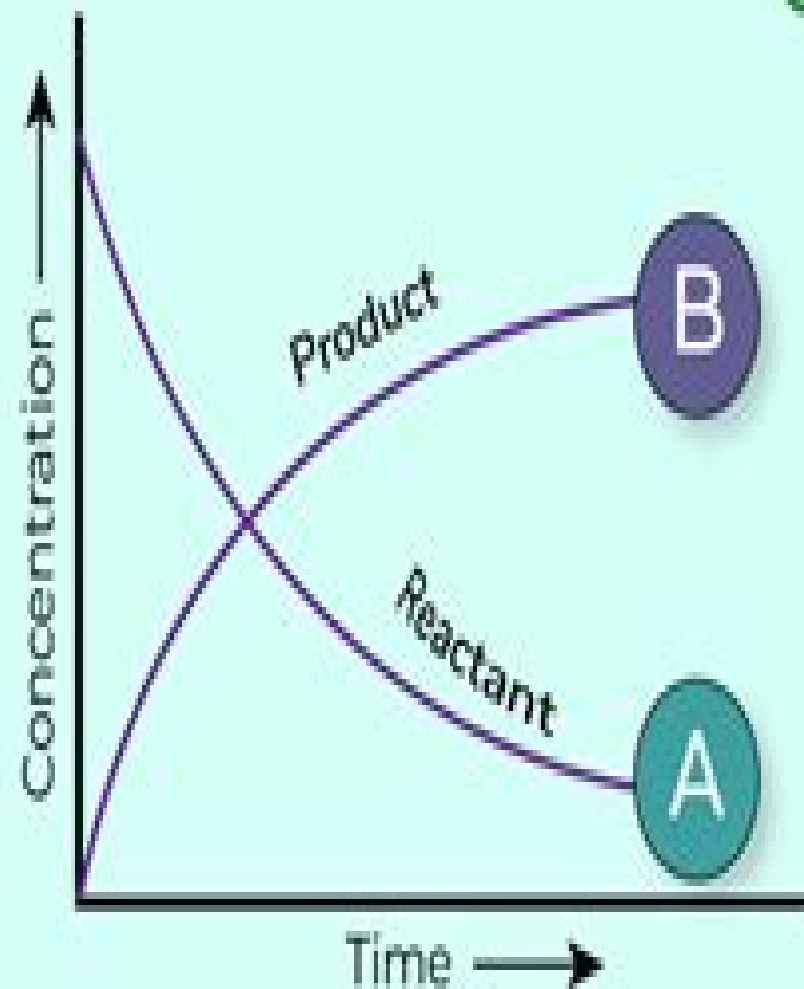


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**National Measurement Laboratory
(U.S.), David Garvin**



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, 1978 Lists citations with abstracts for aerospace related reports obtained from world wide sources and announces documents that have recently been entered into the NASA Scientific and Technical Information Database Chemical Kinetics Luis Arnaut, 2021-06-22 Chemical Kinetics From Molecular Structure to Chemical Reactivity Second Edition explains how molecular structures change with time It offers a comprehensive and coherent coverage of the rates of chemical transformations The book is written for both undergraduate chemistry students and for the specialist The newcomer will find the fundamental concepts the simple experiments and the underlying theories For the seasoned specialist it presents sophisticated experimental and theoretical methods offering a panorama of time dependent molecular phenomena connected

by a new rationale The gap between the two is bridged by a logical path that leads the reader from a phenomenological approach of molecular changes to the formalism of chemical reaction rates and then to state of the art calculations of rate constants of the most prevalent reactions atom transfers catalysis proton transfers substitution reactions energy transfers and electron transfers In the process the reader is presented with the details of collision and transition state theories The coverage includes unimolecular reactions in the gas phase reactions in solution and reactions on surfaces All first edition chapters were revised and most were extended Features two new chapters one on Pharmacokinetics and the other on Oscillatory Reactions and Chaos Includes practical examples detailed theoretical calculations and cross relations between reactions throughout the text to underscore key concepts The rigor of mathematical description of phenomena is combined with simple and profusely illustrated concepts Provides a state of the art presentation on the kinetics of reactions implicated in the most active research fields *Energy Research Abstracts* ,1992 U.S. Government Research Reports ,1963

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the theory of the activated complex in gas and condensed phases Three chapters are devoted to the particular areas of chemical reactions chain reactions catalysis and the stoichiometric heterogeneous reactions Finally the non steady state processes of combustion and explosion are treated in the final chapter *A Computational Study of the Chemical Kinetics of Hydrogen Combustion* T. L. Burks, 1981 A set of elementary reactions and their corresponding rate coefficients has been assembled to describe the homogeneous H_2 O_2 reaction system over the temperature range 300–3000 K The reaction mechanism was drawn together assuming that H_2 O_2 reactive mixtures could be adequately described in terms of self consistent thermal distributions of electronically neutral ground state reactants intermediates and products The resulting time dependent ordinary differential equations describing the system were integrated assuming various initial pressures temperatures and initial concentrations of reactants and diluents The computed results have been compared with experimentally observed induction times second explosion limits the rate of reaction above the second explosion limit and the temporal behavior of reaction species The good agreement between the computational and experimental results attests to the accuracy of the assembled mechanism in its description of the homogeneous reaction system and supports the validity of the set of associated rate coefficients for the elementary reactions of the mechanism over a broad range of reaction conditions Author

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