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[\$\text{hno}_3 + \text{mg} + \text{c}_2\text{h}_3\text{o}_2 \rightarrow 2\text{hc}_2\text{h}_3\text{o}_2 + \text{mg}^{2+} + \text{no}_3^-\$ chemical equation](#) - Aug 11 2023

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each compound to balance $\text{Mg} + \text{HNO}_3 \rightarrow \text{Mg}(\text{NO}_3)_2 + \text{NO}_2 + \text{H}_2\text{O}$ you'll need to be

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web magnesium hydroxide nitric acid balanced total equation $\text{Mg}(\text{OH})_2(\text{aq}) + 2\text{HNO}_3(\text{aq}) \rightarrow \text{Mg}(\text{NO}_3)_2(\text{aq}) + 2\text{H}_2\text{O}$ total ionic equation

$\text{Mg}^{2+}(\text{aq}) + 2\text{OH}^{-}(\text{aq}) + 2\text{H}^{+}(\text{aq}) + 2\text{NO}_3^{-}(\text{aq}) \rightarrow \text{Mg}^{2+}(\text{aq}) + 2\text{NO}_3^{-}(\text{aq}) + 2\text{H}_2\text{O}(\text{l})$ net ionic equation $2\text{OH}^{-}(\text{aq}) + 2\text{H}^{+}(\text{aq}) \rightarrow 2\text{H}_2\text{O}(\text{l})$ sulfuric acid and lithium hydroxide

balanced total equation $\text{H}_2\text{SO}_4(\text{aq}) + 2\text{LiOH}(\text{aq}) \rightarrow \text{Li}_2\text{SO}_4(\text{aq}) + 2\text{H}_2\text{O}(\text{l})$