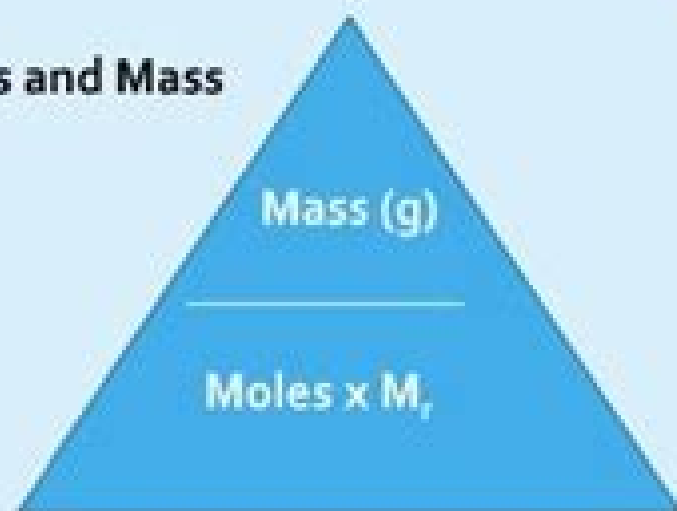




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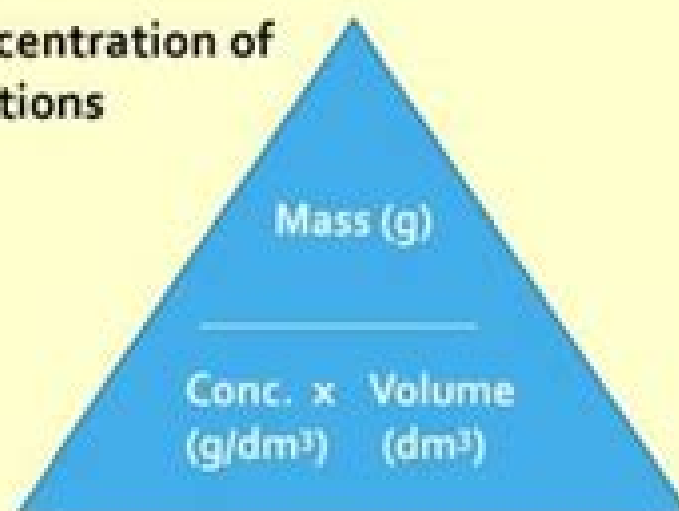
## Topic 3 – Quantitative Chemistry

### Moles and Mass



$M_r$  (relative formula mass) = sum of relative atomic mass in quantities given in the formula e.g.  $\text{Cl}_2 = 35.5 \times 2$

### Concentration of solutions



### Volumes gases

$$\% \text{Yield} = \frac{\text{Amount of product produced (g)}}{\text{Max. amount of product possible (g)}} \times 100\%$$

$$\text{Atom economy} = \frac{M_r \text{ desired product from the reaction}}{\text{Sum of } M_r \text{ of all reactants}} \times 100\%$$

# Quantitative Chemistry

**M Walker**



## Quantitative Chemistry:

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