

Compd	Mp (°C) ^a	Yield (%)	Formula
2a	169–172	52	C ₁₀ H ₆ FNO ₂ S
2b	178–180	56	C ₁₁ H ₉ NO ₂ S
2c	146–148	65	C ₁₆ H ₁₁ NO ₃ S
2d	187–190	52	C ₁₁ H ₉ NO ₃ S
2e	180–182	57	C ₁₁ H ₆ F ₃ NO ₂ S
2f	263–265	88	C ₁₇ H ₁₂ N ₂ O ₃ S
2g	210–212	65	C ₁₀ H ₆ FNO ₂ S
2h	183–186	55	C ₁₆ H ₁₁ NO ₃ S
2i	216–218	70	C ₁₁ H ₉ NO ₃ S
2j	206–208	54	C ₁₁ H ₆ F ₃ NO ₂ S
2k	288–289	90	C ₁₇ H ₁₂ N ₂ O ₃ S
3a	112–115	96	C ₁₃ H ₁₀ FNO ₄ S
3b	92–95	78	C ₁₄ H ₁₃ NO ₄ S
3c	117–120	76	C ₁₉ H ₁₅ NO ₅ S
3d	48–50	98	C ₁₄ H ₁₃ NO ₅ S
3e	170–173	88	C ₁₄ H ₁₀ F ₃ NO ₄ S
3g	221–223	78	C ₁₃ H ₁₀ FNO ₄ S
3j	192–195	89	C ₁₄ H ₁₀ F ₃ NO ₄ S
4a	215–216	96	C ₁₂ H ₈ FNO ₄ S
4b	230–231	98	C ₁₃ H ₁₁ NO ₄ S
4c	216–217	94	C ₁₈ H ₁₃ NO ₅ S
4d	153–155	98	C ₁₃ H ₁₁ NO ₅ S
4e	213–215	98	C ₁₃ H ₈ F ₃ NO ₄ S
4g	240–245	96	C ₁₂ H ₈ FNO ₄ S
	(dec.)		
4j	251–252	97	C ₁₃ H ₈ F ₃ NO ₄ S

^aRecrystallization solvent: MeOH for compounds 2 (EtOH/H₂O for 2f); EtOH/H₂O for 3; EtOH for 4.

Macmillans Chemical Physical Data

Peter Atkins,Julio de Paula



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Macmillan's Chemical and Physical Data Arthur M. James, Mary P. Lord, 1992 This handbook provides data on the more important properties of the most widely used compounds in 15 defined subject areas Given in SI units the data is presented in some 430 tables most of which are provided with references for further information and or additional data Special features include emphasis on growth points such as superconductivity lasers new standards and toxicity and recommended exposure limits for hazardous chemicals

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