

Quantitative Analysis

Definition, Methods, Advantages



1. Weighing the sample
of the analyte

2. Dissolving the
sample in water

3. Adding a reagent
to form a precipitate

4. Filtering to collect
the precipitate

5. Drying the precipitate
to constant weight
to determine the amount of
analyte present

Gravimetric analysis



1. Preparing the standard
solution

2. Standardizing the
solution

3. Titrating the sample

4. Reaching the endpoint
color change

Volumetric analysis

Principles Of Quantitative Chemical Analysis

Charles William Foulk



Principles Of Quantitative Chemical Analysis:

Principles of Quantitative Chemical Analysis Robert De Levie,1997 Designed for a sophomore junior course in analytical chemistry or quantitative analysis this text focuses on the quantitative aspects of the discipline using a unified approach Emphasis is placed on developing visual tools for understanding complicated solution equilibria To these ends extensive use is made of graphical methods such as the easily sketched stick diagrams which can be used to guide analytical calculations and takes the guesswork out of numerical approximations Optional spreadsheet exercises are closely integrated with the text and can therefore serve to introduce the student to the use of computers for chemical calculations

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