Molecular mechanics:

- The "mechanical" model developed to describe molecular structures and properties in as practical manner as possible
- Molecular mechanics methods are based on the following principles:
- Nuclei and electrons are lumped into atom-like particles.
- Atom-like particles have a net charge
- Interactions are based on springs and classical potentials.
- Interactions must be pre-assigned to specific sets of atoms.
- Interactions determine the spatial distribution of atom-like particles and their energies







Principles Of Molecular Mechanics

Olivier Darrigol

Principles Of Molecular Mechanics:

Principles of Molecular Mechanics Katsunosuke Machida,1999-09-10 Principles of Molecular Mechanics Katsunosuke Machida Computational methods in chemistry have become increasingly important over recent years and today many chemical laboratories in industry and academia are routinely applying the principles of molecular mechanics. This unique book written from a theoretical chemist's point of view brings together the mathematical and theoretical basis of calculations used in many molecular mechanics software tools and will be indispensable for anyone using computational techniques. Principles of Molecular Mechanics contains a discussion of the fundamental analytical expressions used in calculating molecular properties from molecular force fields derived from a wide variety of mathematical and physical methods Practical algorithms are outlined with an emphasis on speeding up calculation and saving computer memory essential to researchers designing or improving computer programs for molecular mechanics. This book is essential reading for all researchers and graduate students working in molecular simulations computational chemistry theoretical chemistry and physical chemistry

Fundamental Principles of Molecular Modeling Anton Amann, Jan C.A. Boeyens, W. Gans, 2013-06-29 Molecular similarity has always been an important conceptual tool of chemists yet systematic approaches to molecular similarity problems have only recently been recognized as a major contributor to our understanding of molecular properties Advanced approaches to molecular similarity analysis have their foundation in quantum similarity measures and are important direct or indirect contributors to some of the predictive theoretical computational and also experimental methods of modern chemistry This volume provides a survey of the foundations and the contemporary mathematical and computational methodologies of molecular similarity approaches where special emphasis is given to applications of similarity studies to a range of practical and industrially significant fields such as pharmaceutical drug design The authors of individual chapters are leading experts in various sub fields of molecular similarity analysis and the related fundamental theoretical chemistry topics as well as the relevant computational and experimental methodologies Whereas in each chapter the emphasis is placed on a different area nevertheless the overall coverage and the wide scope of the book provides the reader with a general yet sufficiently detailed description that may serve as a good starting point for new studies and applications of molecular similarity approaches The editors of this volume are grateful to the authors for their contributions and hope that the readers will find this book a useful and motivating source of information in the rapidly growing field of molecular similarity analysis Physics and Necessity Olivier Darrigol, 2014 This book recounts a few ingenious attempts to derive physical theories by reason only beginning with Descartes geometric construction of the world and finishing with recent derivations of quantum mechanics from natural Principles and Applications of Quantum Chemistry V.P. Gupta, 2015-10-15 Principles and Applications of axioms Quantum Chemistry offers clear and simple coverage based on the author's extensive teaching at advanced universities around the globe Where needed derivations are detailed in an easy to follow manner so that you will understand the physical

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