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# Seventeen And In Between

**Eugenio Garin**

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web a reaction and its experimental rate law are represented below  $2 \text{ HCl} + \text{H}_2\text{S} \rightarrow \text{S} + 2 \text{ H}_2\text{O}$   $\text{HCl}$  rate  $k_1 \text{ HCl}$   $k_2 \text{ H}_2\text{S}$  a chemist

proposes a mechanism for the reaction that is consistent with the rate law the mechanism has two elementary steps and the first step is slow compared to the second

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